

UnLOHCked

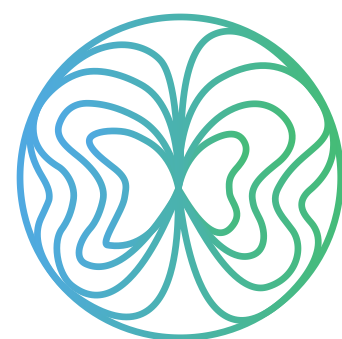
UNlocking the potential of LOHCs through the development of KEy sustainable and efficient systems for Dehydrogenation

HORIZON-JTI-CLEANH2-2022-02-05

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IRCELYON - CP2M

May 2023



Ircelyon

INSTITUT DE RECHERCHES
SUR LA CATALYSE
ET L'ENVIRONNEMENT

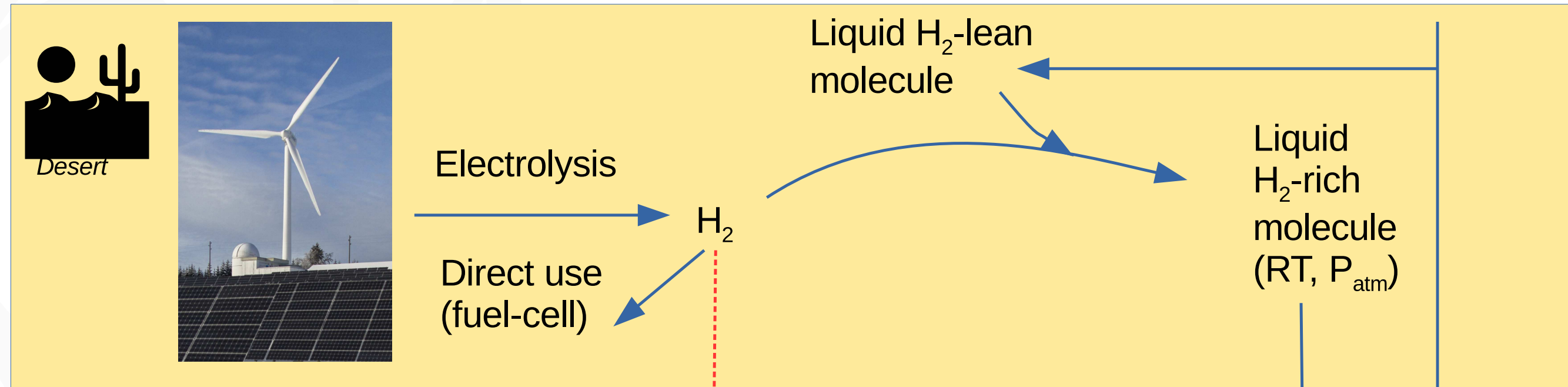


Context of LOHC

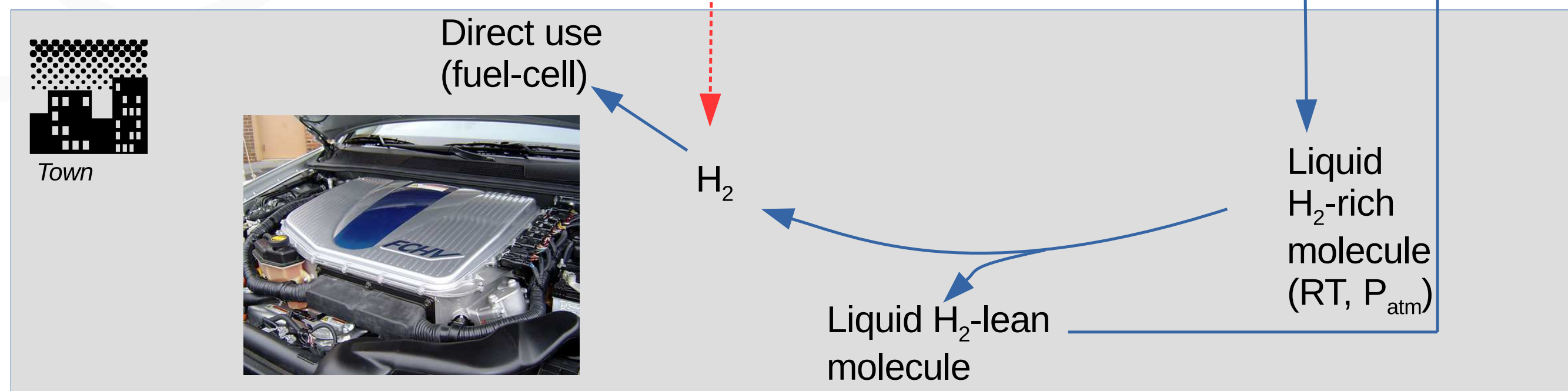
Context of LOHC

- The project
- Methodology
- Tasks
- Dehydrogenation issues
- Dehydrogenation reactor
- Liquid phase analysis
- H0-BT/H12-BT pair
- Catalysts
- Bimetallic catalysts
- Single-atom catalysts...
...or nano-clusters
- Hsp supports for hydrogenation
- Hsp supports for dehydrogenation
- Conclusion
- ICC 2024

Upstream coupling (electrolysis + hydrogenation)



Downstream coupling (dehydrogenation + FC)



 **Costly and unsafe storage and transportation**

Easy and safe storage and transportation





The project

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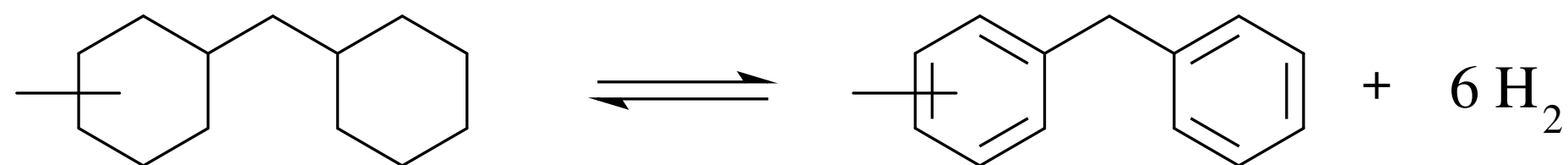
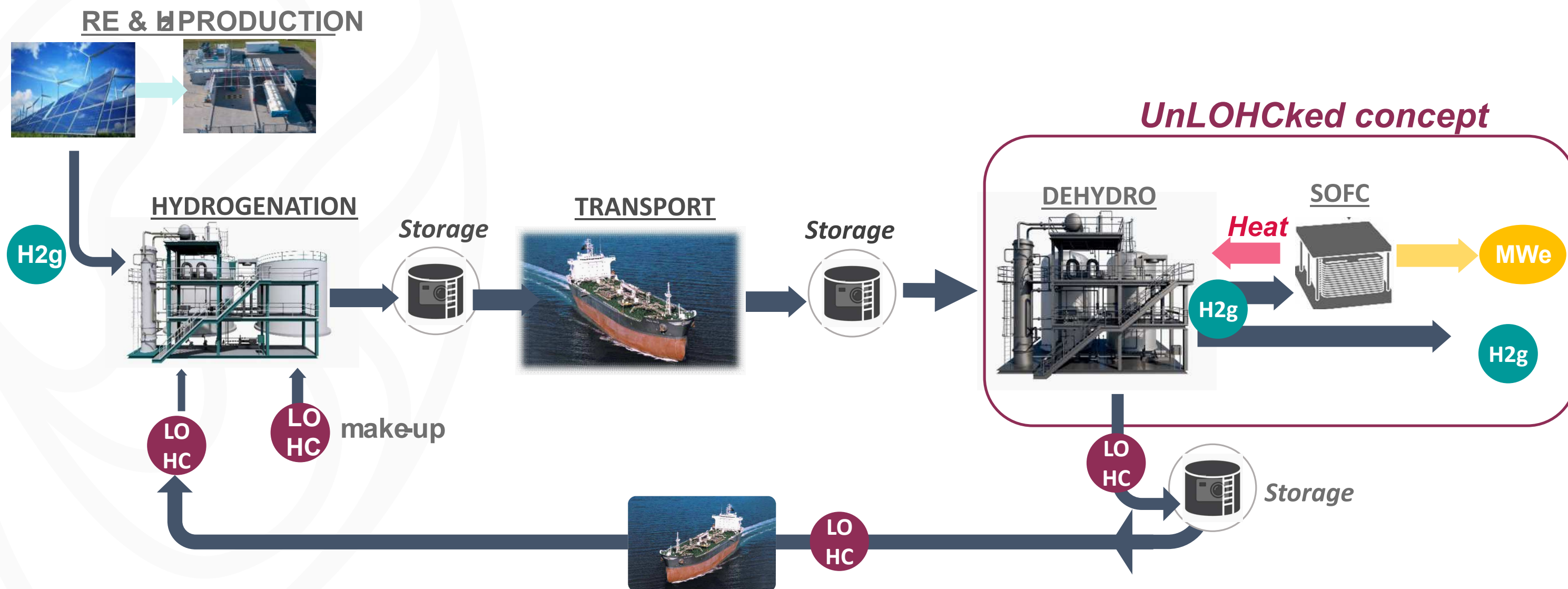
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Hsp supports for hydrogenation

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H12-BT/H0-BT pair

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EXPECTED OUTCOMES

CATALYST DEVELOPMENT

Development of low/free CRM catalyst for LOHC dehydrogenation at lower temperature

TRL3



- Novel catalysts with metallic/bimetallic structures
- CRM-free or low CRM content
- Novel preparation processes in order to increase metal dispersion in an engineered support

SYSTEM VALIDATION

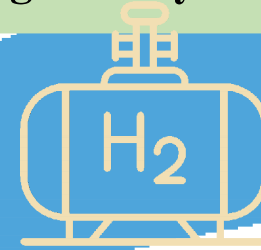
Testing and evaluation in stationary unit



- Batch & continuous testing

REACTOR DESIGN & MANUFACTURE

Conversion >95%; improved thermal efficiency; cost effective & ease of operation; integrated systems



- Modelling activities
- Thermal coupling for the system integration: dehyd reactor & SOFC

UPSCALING

Demo system, running for ≥ 500 hours and producing ≥ 10 kg H₂/day at atmospheric pressure



TRL5

- CO₂-free dehyd system integrated: SOFC & LOHC operating

LARGE SCALE

Scalability of the developed system to large-scale production for long distance transportation

- Life-cycle assessment (LCA) for the whole supply chain
- Techno-economic analysis (TEA) for 100-1,000 t H₂/day system

ACTIVITIES & CHALLENGES

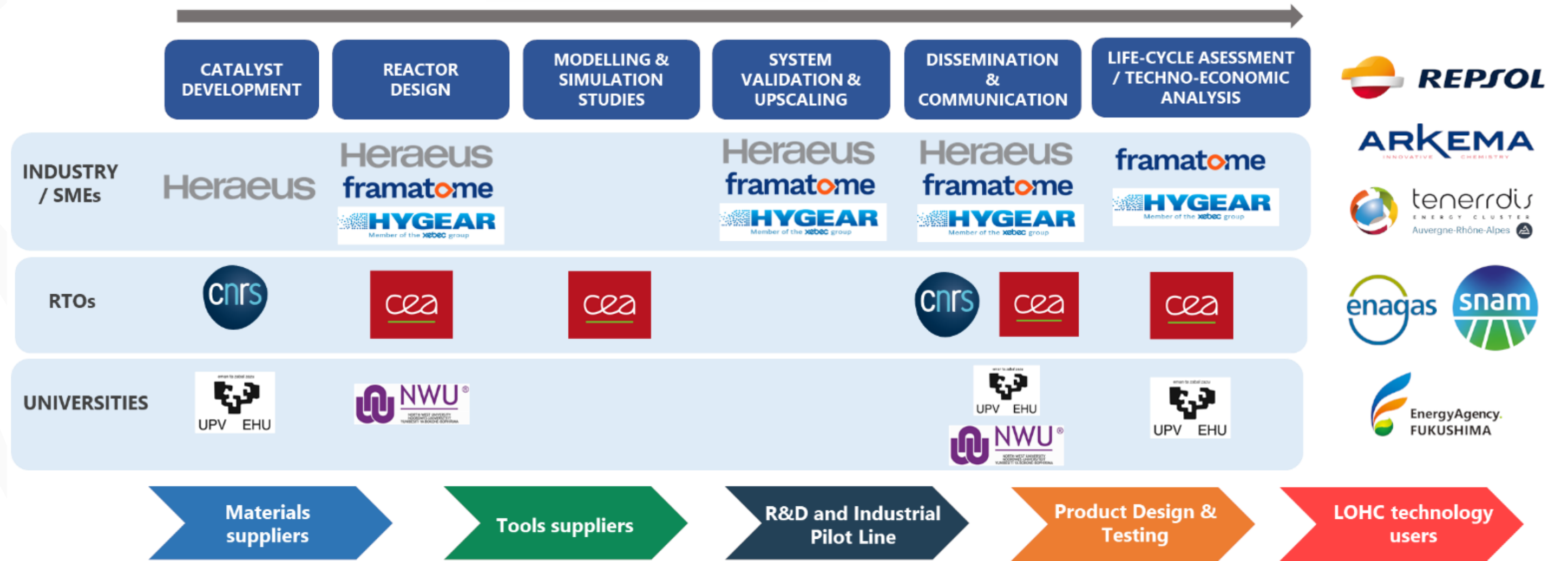
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- Only PGM catalysts are highly active and selective
- The dehydrogenation enthalpy is high and the reaction must be performed at high $T^{\circ}\text{C}$ (250-300 $^{\circ}\text{C}$)
- Some by-products of the reaction (traces) avoid the complete cycling dehydrogenation/hydrogenation
- Some catalyst deactivation may occur

Dehydrogenation reactor

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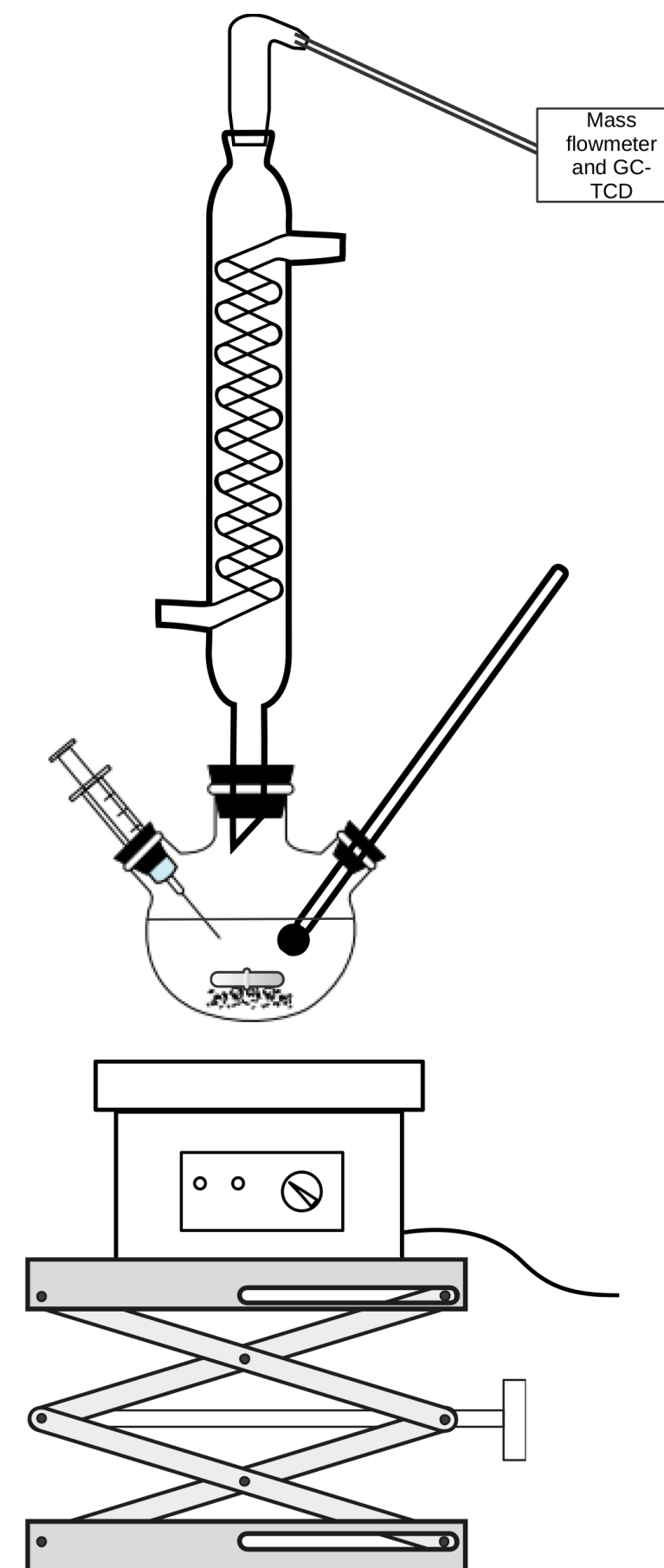
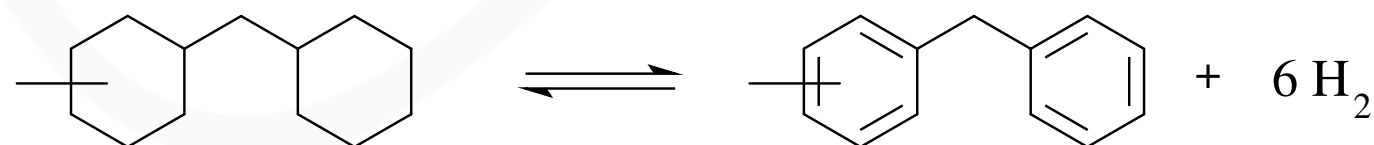
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Test in stirred reactors (G/L/S) with continuous removal of hydrogen (H₂ flow rate measurement).

Different temperatures

Different catalyst loadings

No solvent



Liquid phase analysis

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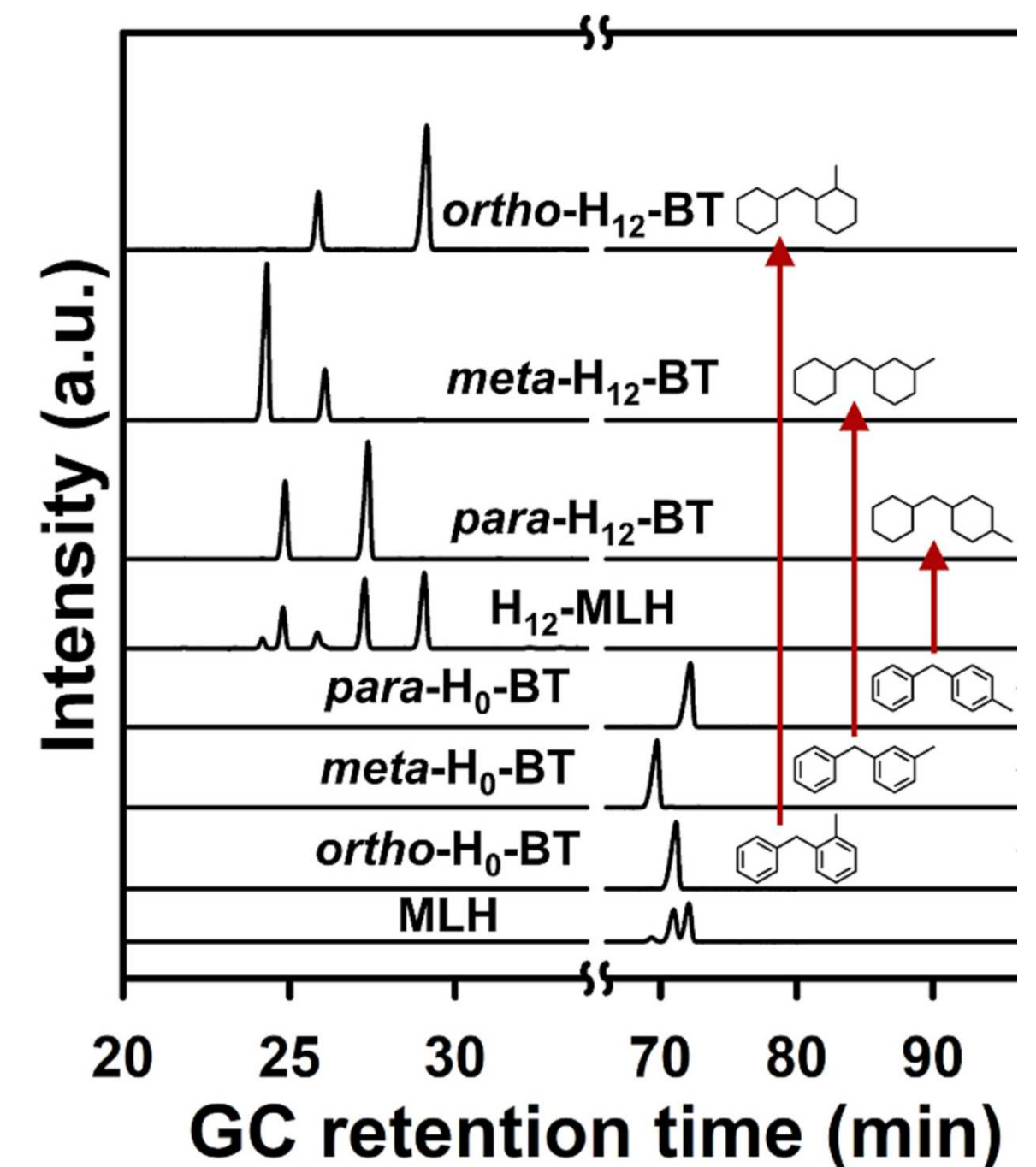
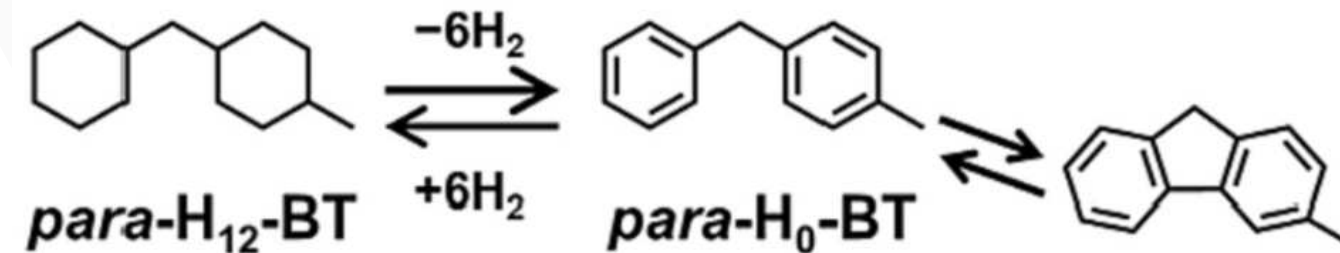
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Analysis of LOHC+ and LOHC-
molecules by GC

BUT

many isomers at each
dehydrogenation degree.

Known by-product:
methyl-fluorene

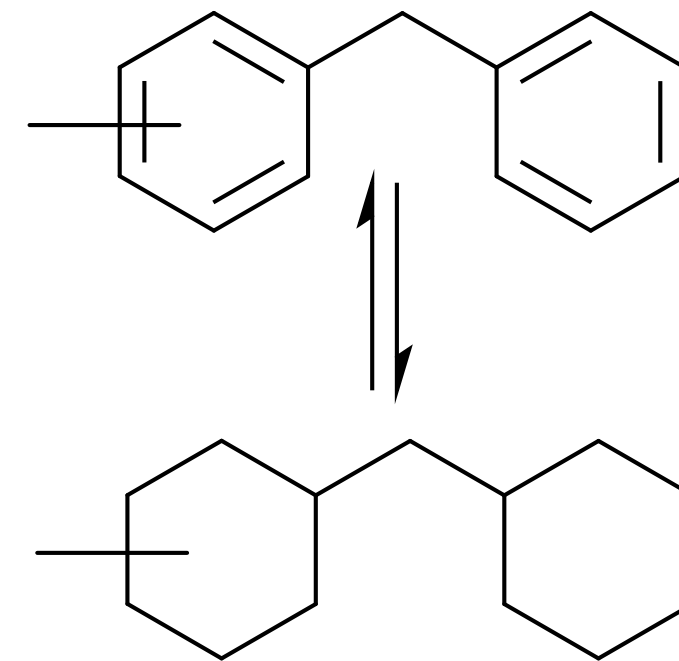
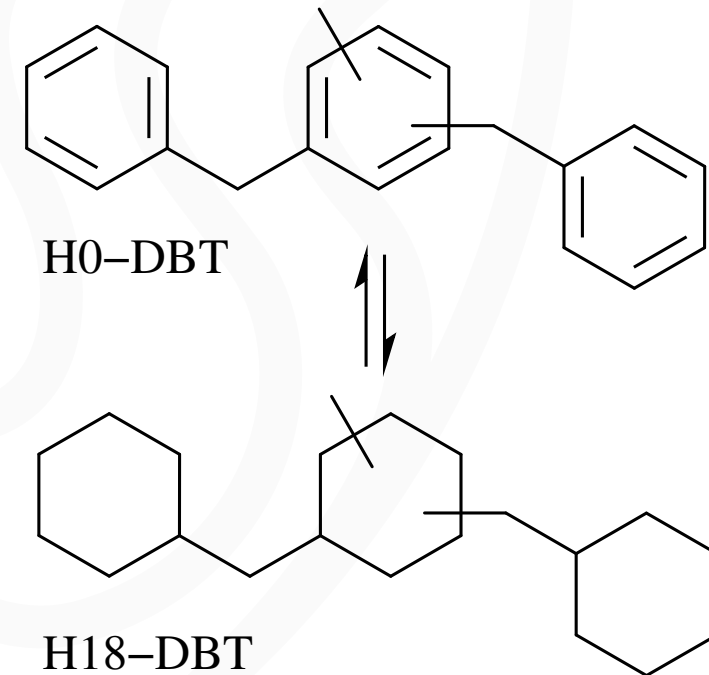


H0-BT/H12-BT pair

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Few developments dedicated to Benzyltoluene/H12-BT.

System much less studied than H0-DBT/H18-DBT, should be less complex.



- High volumetric storage density ($54.5 \text{ kgH}_2/\text{m}^3$ at RT)
- Liquid in a large range of $T^\circ\text{C}$ (-70 to +270 $^\circ\text{C}$)
- Excellent robustness in hydrogenation/dehydrogenation cycles
- Low viscosity for easy handling
- Reaction rates (Hyd/Dehyd) higher compared to those of the DBT-based LOHC system (same conditions).



Catalysts

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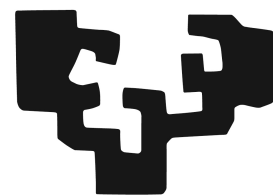
Some ideas in the literature:

- WO_x -incorporated $\text{Pt}/\text{Al}_2\text{O}_3 > \text{Pt}/\text{Al}_2\text{O}_3$ (H18-DBT)
C. H. Kim et al. Fuel, 2022, 313, 122654
- Ni-modified catalysts active (MCH)
Y. K. Gulyaeva et al. Catalysts, 2020, 10, 1198

Strategy followed for the UnLOHCked project:

- Develop bimetallic catalysts (NiZn, NiCu, NiSn but also PtSn, PtNi)
- Develop single-atoms catalysts
- Develop hydrogen spillover (Hsp) zeolites
- Re, Fe, Mo, Zr mono and bimetallic catalysts (with Pt). Conventional preparation (IWI) and ScCO_2 preparation

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UPV EHU

Bimetallic catalysts

NiCu and NiZn known to selectively dehydrogenate methylcyclohexane.
2nd metal allows an increase of selectivity compared to Ni alone.

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Bimetallic catalysts

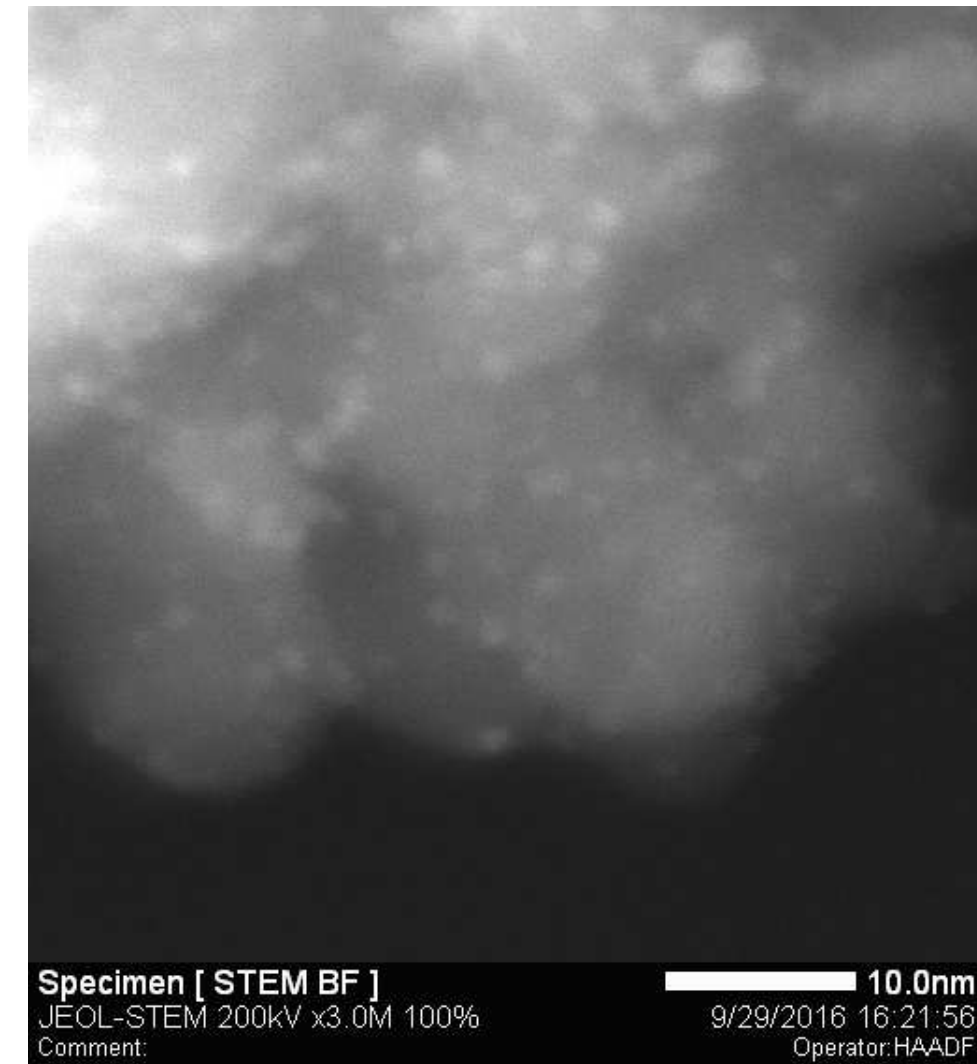
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NiCu and NiZn known to selectively dehydrogenate methylcyclohexane.

2nd metal allows an increase of selectivity compared to Ni alone.

Strategy:

- co-impregnation
- 2-steps SOMC chemistry
- Preparation of Ni silicide NPs in solution further immobilised onto supports
=> different sizes (from 1.5 to 20-50 nm) and different alloy phases.



STEM-HAADF of Ni/SiO₂

NiZn: Al-ShaikhAli, A. H. et al. Chem. Commun., 2015, 51, 12931-12934

NiCu: Gulyaeva, Y. K. et al. Catalysts, 2020, 10, 1198

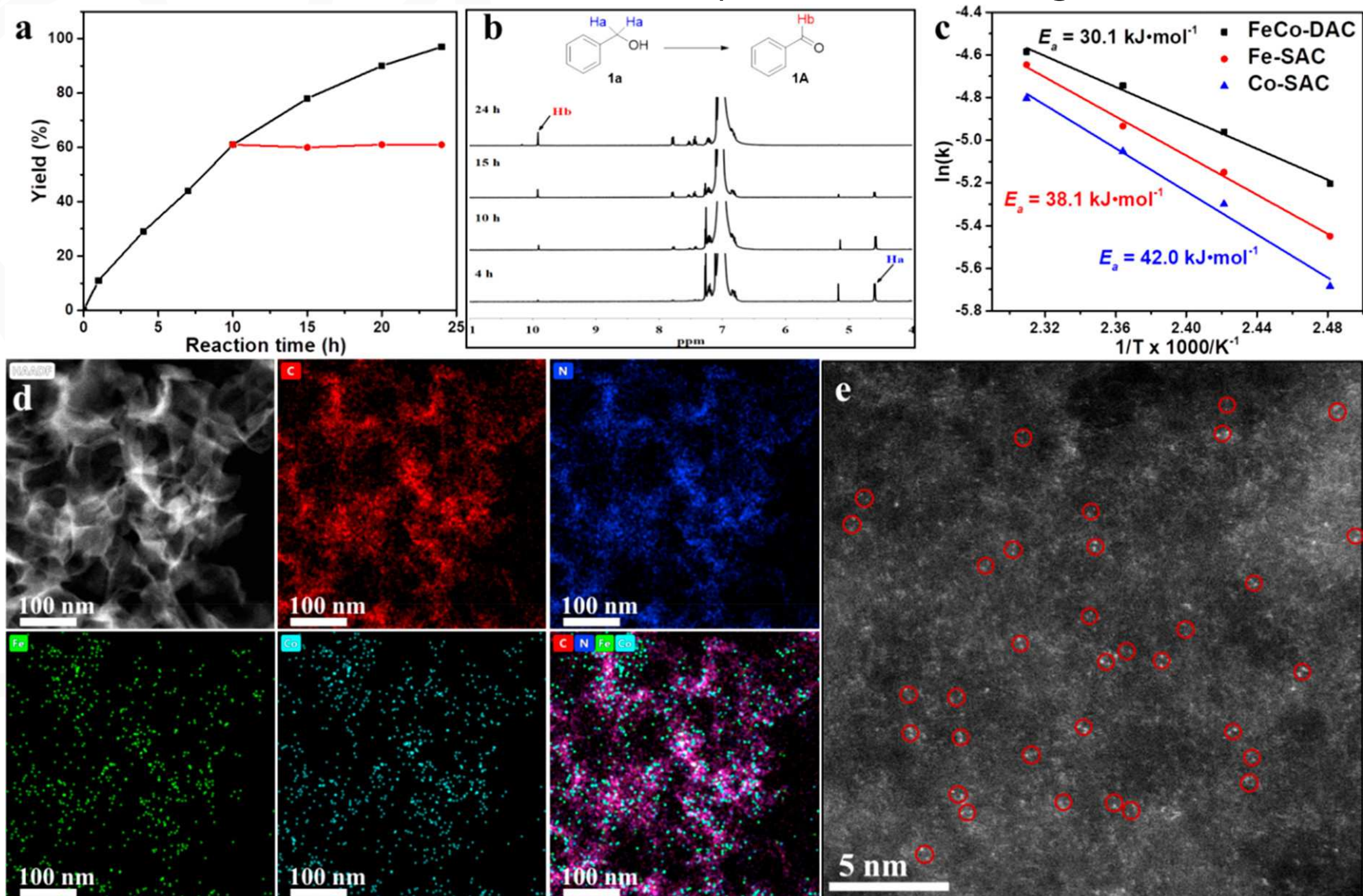
SOMC: Docherty, S. R. et al. JACS Au, 2021, 1, 450-458

Ni NPs: Galeandro-Diamant, T. et al. Catal. Sci. Technol., 2019, 9, 1555 - 1558

Single-atom catalysts...

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Size below the cluster and nanoparticles => single-atom catalysts

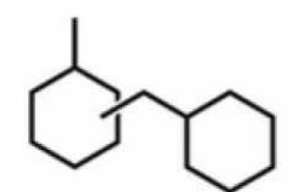


...or nano-clusters

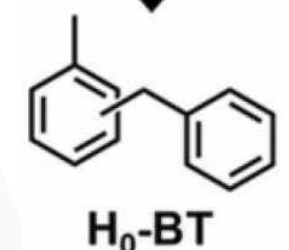
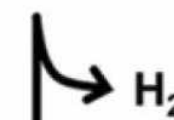
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Mesoporous Pt-0.33MnO_x-Al₂O₃ (mPt0.33MnA)

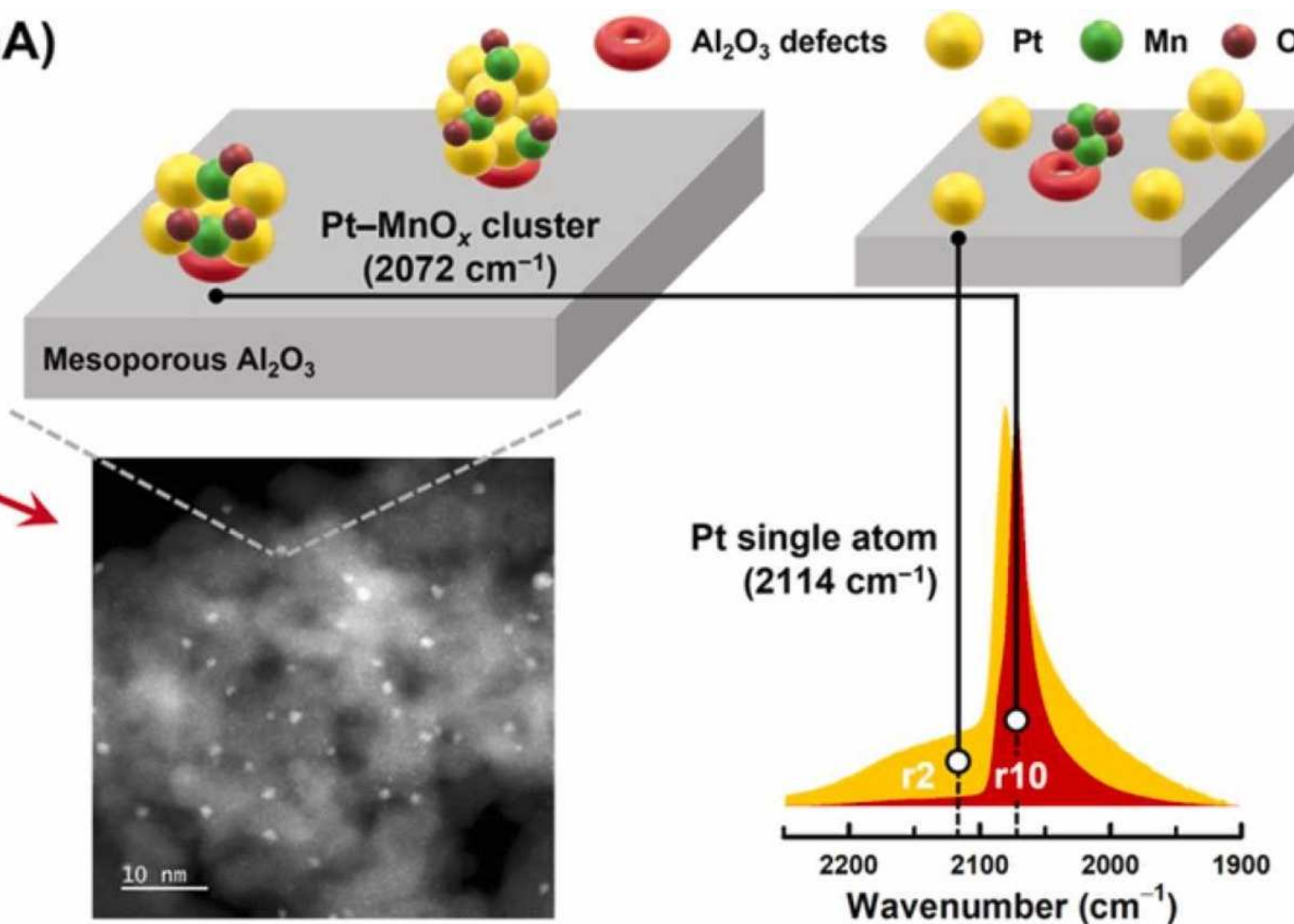
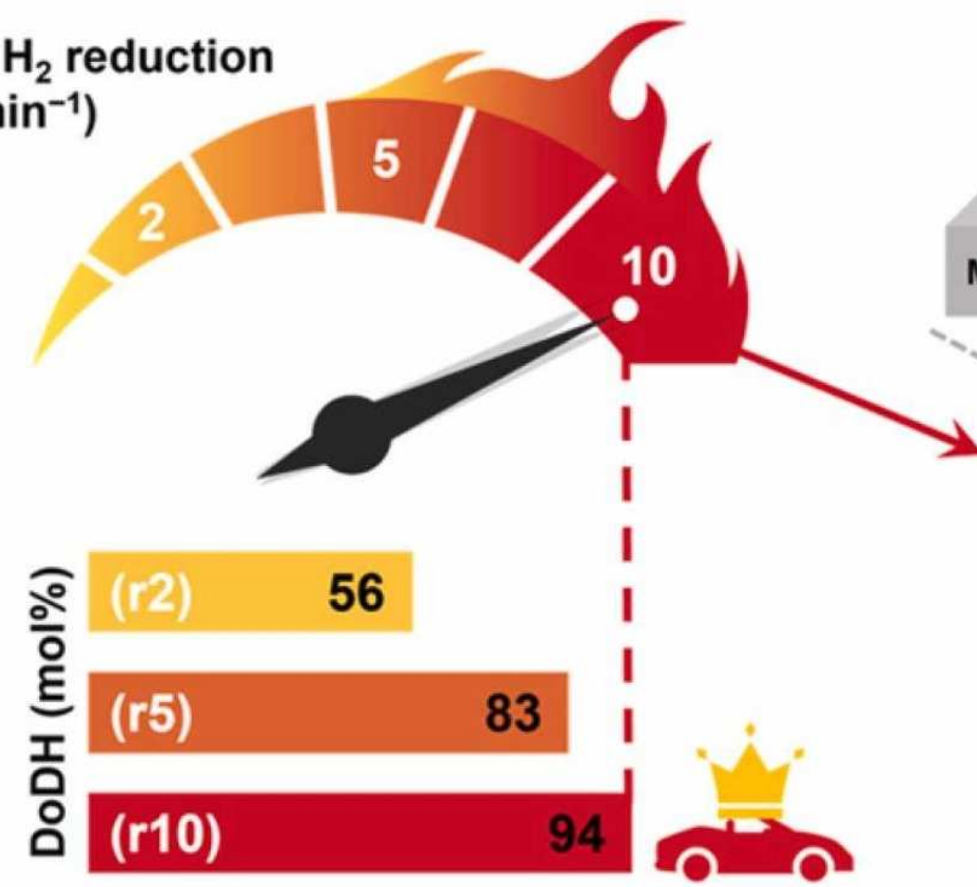
Ramp rate in H₂ reduction
(°C min⁻¹)



H₁₂-BT



H₀-BT



- The higher ramp rate in H₂ reduction of mPtMnA leads to activity improvement.
- Pt nanoclustering with MnO_x (< 2 nm) favored

Jo, Y. et al, *Appl. Catal. B*, 2023, 334, 122848

Hsp supports for hydrogenation

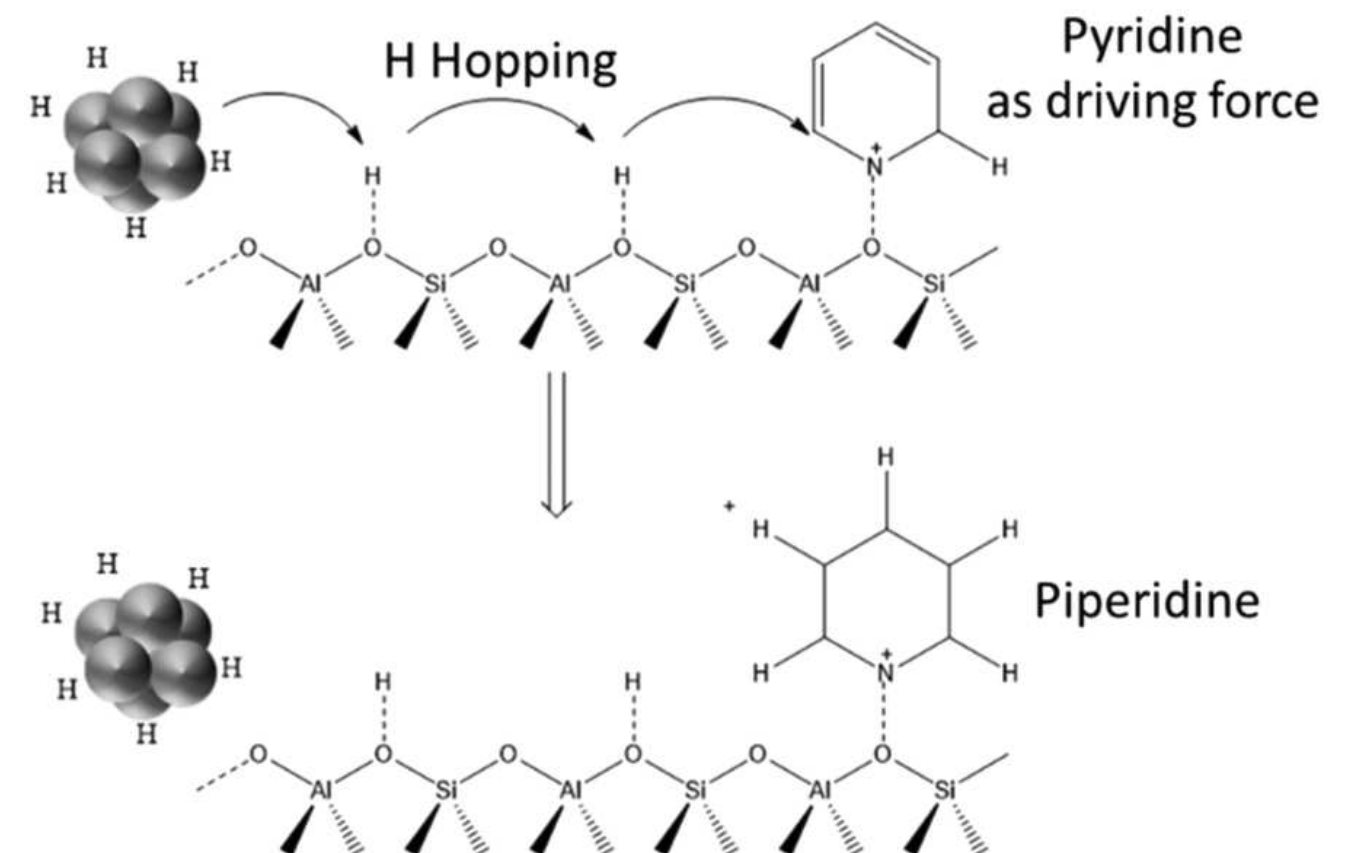
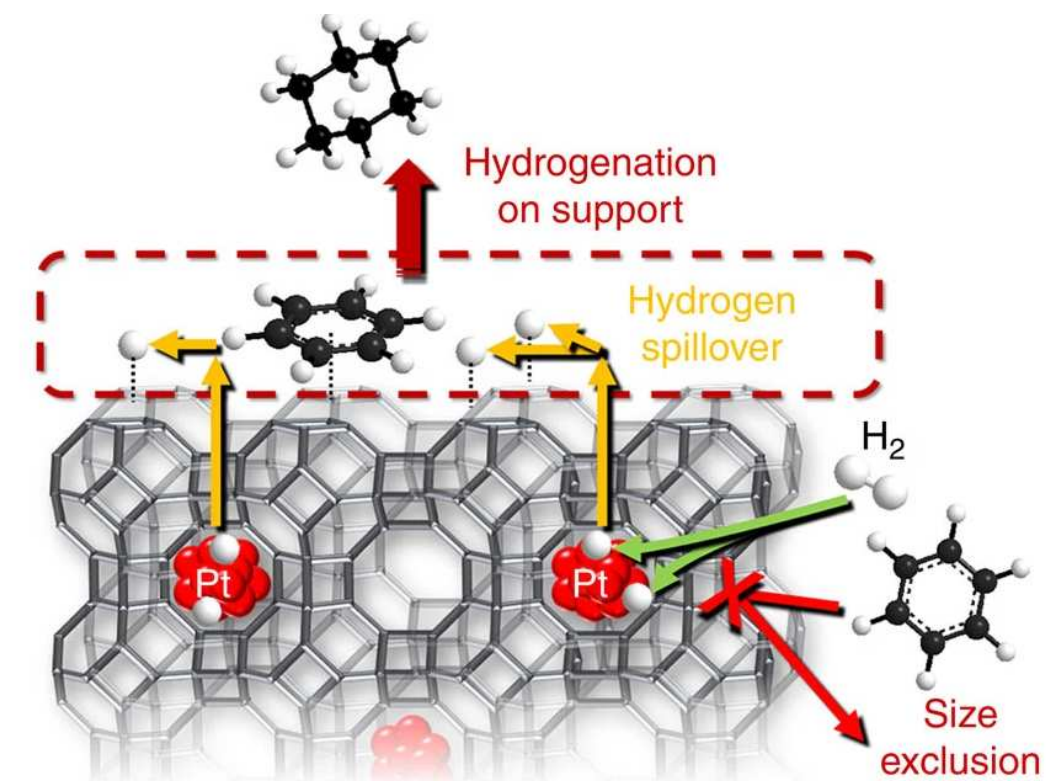
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MFI and CHA, supporting metal nanoclusters (size < 2 nm)
Hydrogen spillover phenomenon to enhance the active phase.

Compromise between H₂ diffusion and H spillover
=> modification of the standard zeolite properties:

- Brønsted acidity
- Si/Al ratio
- Extra framework aluminum
- Si-OH

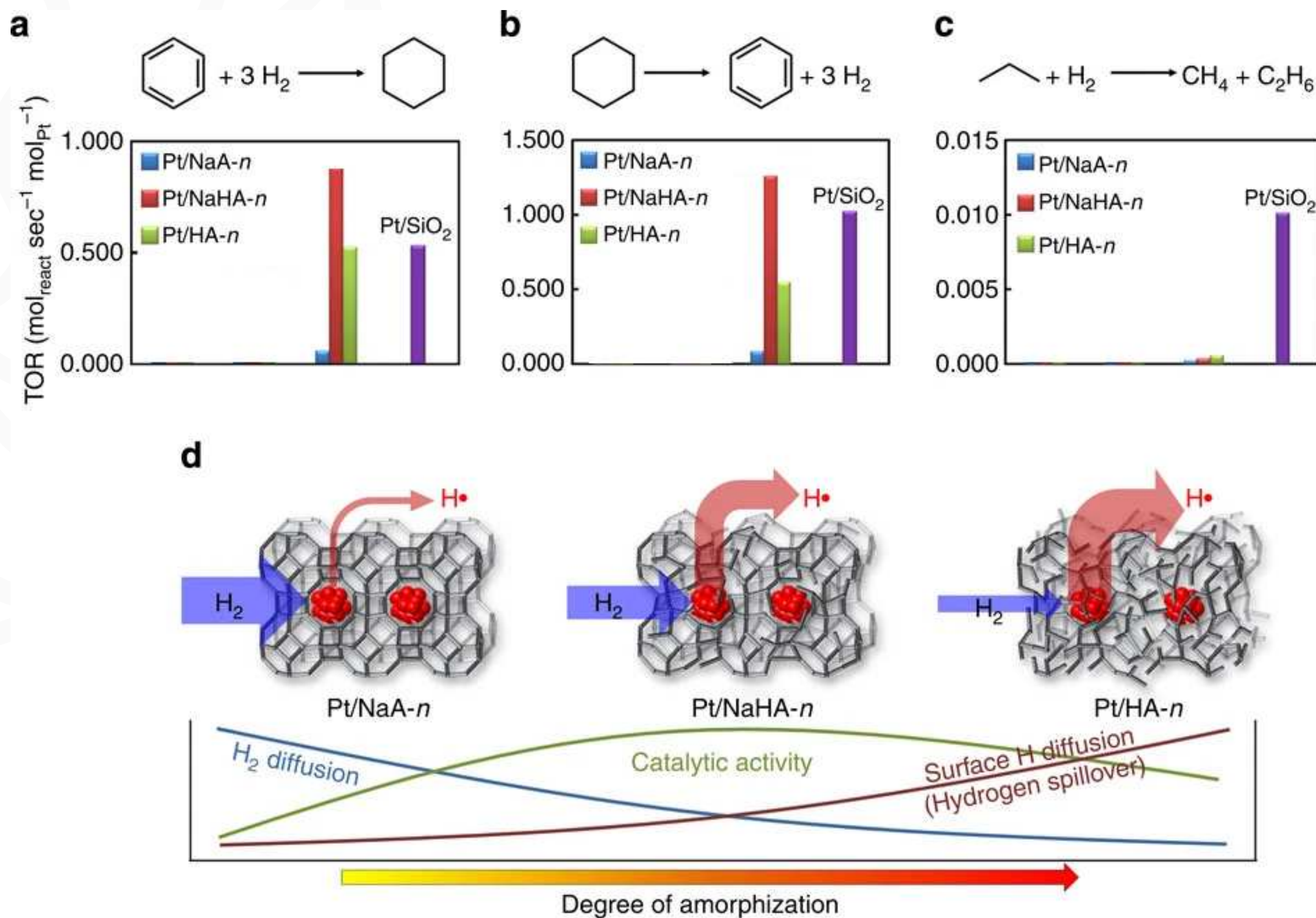
Batalha, N. et al. Catal. Sci. Technol. 2022, 12, 1117-1129
Im, J. et al. Nature Communications, 2014, 5, 3370



Hsp supports for dehydrogenation

Dehydrogenation enhanced by reverse hydrogen spillover

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- UnLOHCkEd kick-off meeting on July
- Catalyst development from July 2023 to July 2025
- Upscaling of the best catalyst(s) by Heraeus
- Coupling with SOFC
- Demo end 2026

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Thanks to:

- Antoinette Boréave
- Marlène Fabre
- Guillaume Fabre
- Franck Morfin
- Laurent Veyre



THANKS FOR YOUR ATTENTION

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Short-symposium "Catalytic reactions for hydrogen storage and release" (R. Janot, N. Bion, V. Meille...).

- LOHC
- Circular storage (NH_3)
- Solid-state H_2 storage: Catalytic reactions to enhance the dehydrogenation

PARTICIPATE TO ICC 2024 in LYON!