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Institut de chimie de la matière condensée de bordeaux

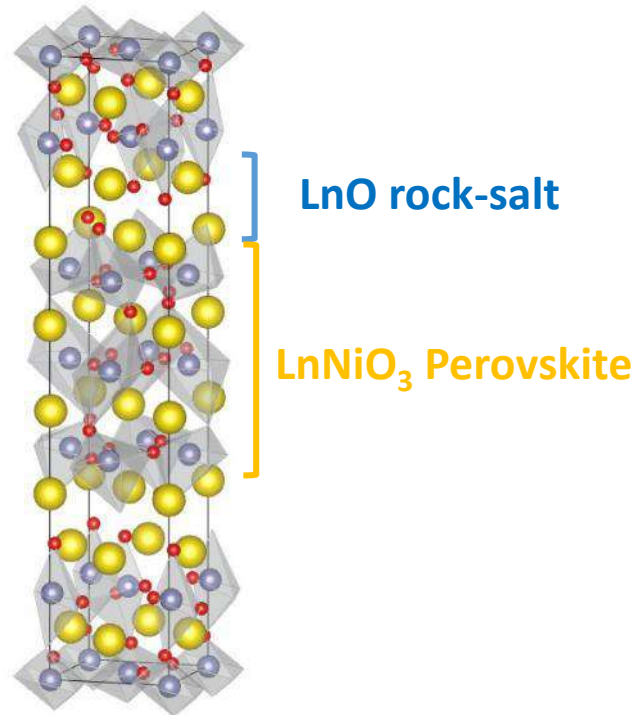
Mixed La/Pr n=3 Ruddlesden-Popper nickelates as stable and efficient oxygen electrodes for high temperature water electrolysis

Presented by: Romuald Frugier, first year Ph.D. student

Co-contributors: Vaibhav Vibhu, Sebastien Fourcade, Jean-Marc Bassat and Jacinthe Gamon

3^e réunion plénière de la fédération hydrogène FRH2 – 23 - 26 mai 2023
Saint-Gilles-Les-Bains – La Réunion

Layered materials: Alternation of n Perovskite layers + 1 rock-salt layer



● Ln (Ln=La, Pr, Nd)

● Ni ● O

Different n-values:

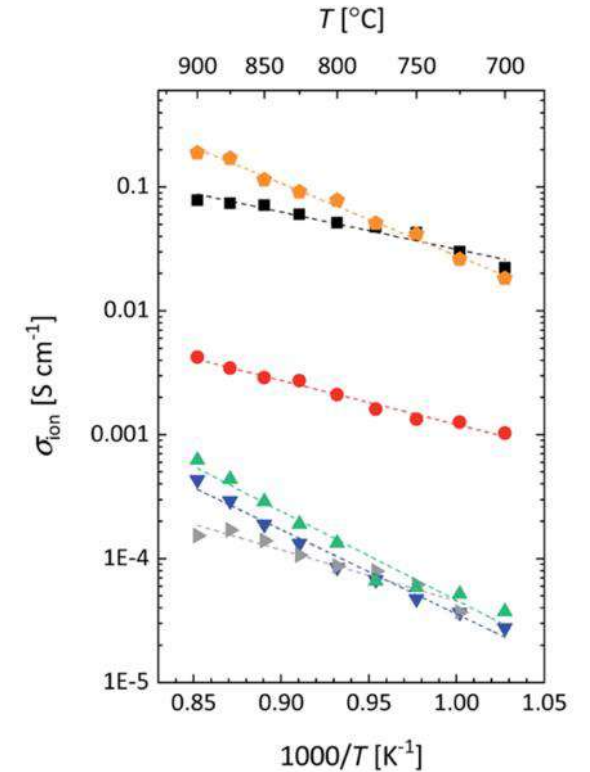
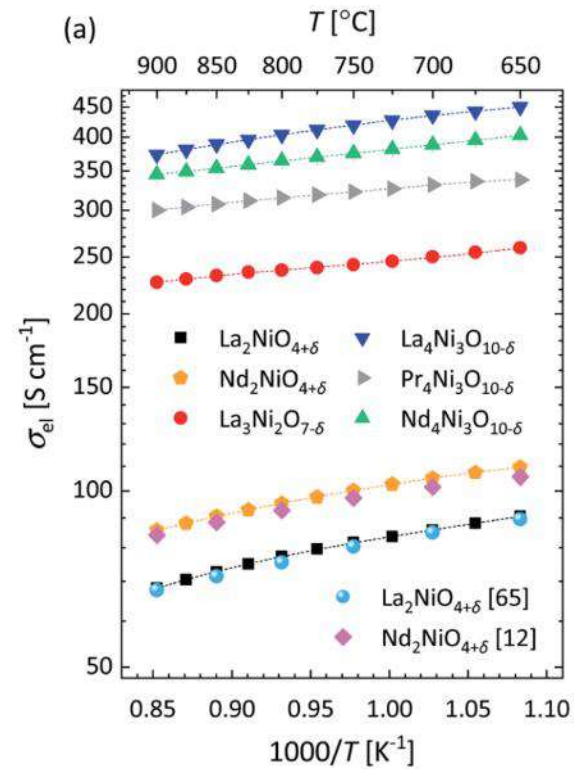
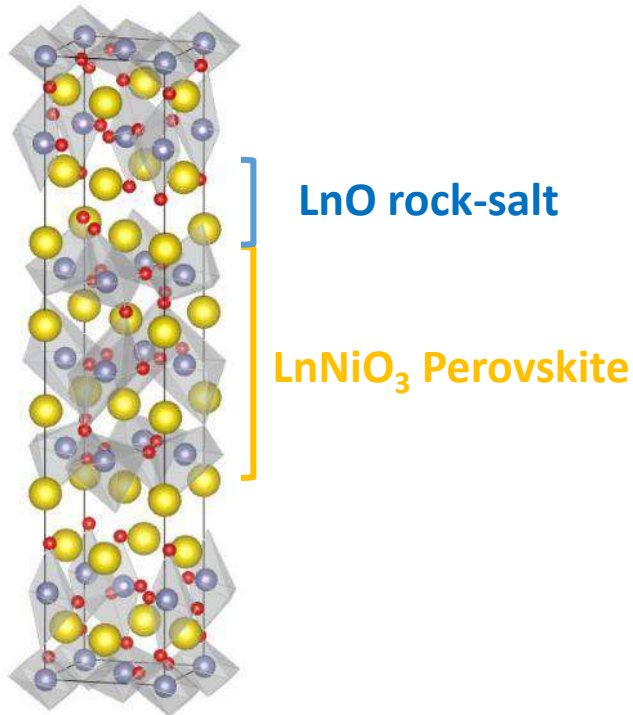
→ n=1 1 perovskite / 1 rock-salt : $\text{Ln}_2\text{NiO}_{4+\delta}$

→ n=3 3 perovskite / 1 rock-salt : $\text{Ln}_4\text{Ni}_3\text{O}_{10\pm\delta}$

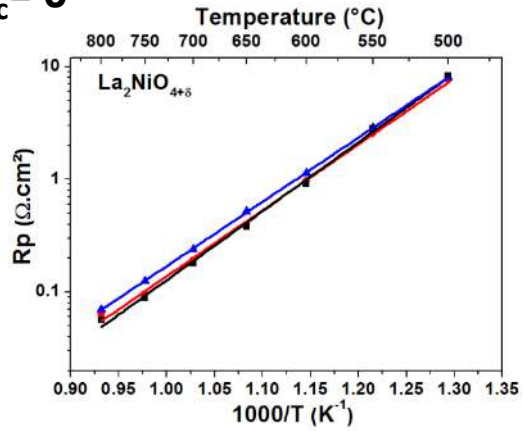
Orthorhombic – Monoclinic systems

Perovskite layers: Electrons conduction pathways

Rock-salt layers: O²⁻ ions conduction pathways



$I_{dc} = 0$



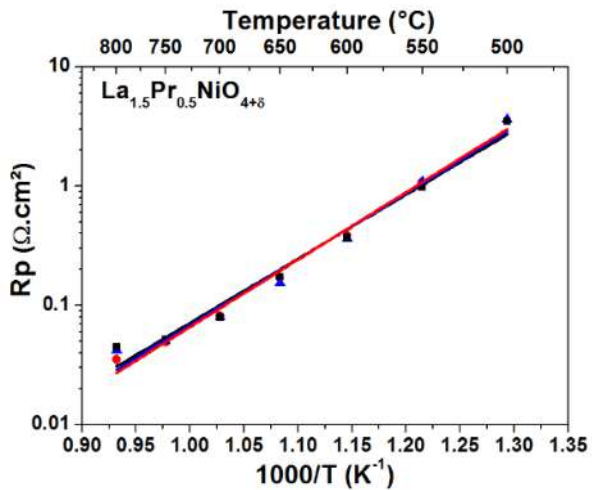
Electrochemical stability

At 700°C

R_p (as prepared) = $0.21 \Omega \cdot \text{cm}^2$

Mixed La/ Pr compound: To combine La-compound thermal stability and Pr-compound high performances.

$$I_{dc} = 0$$



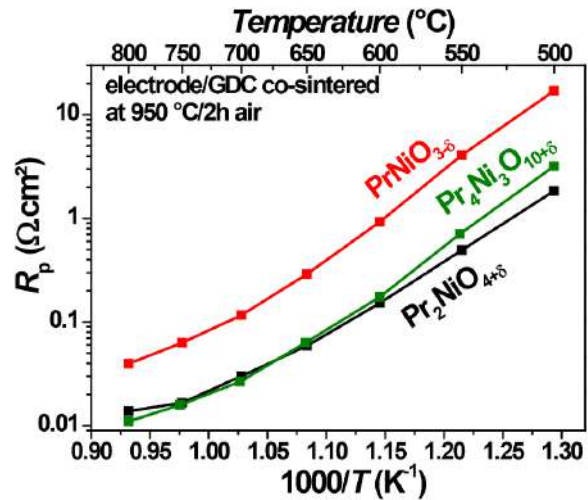
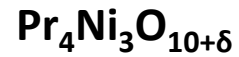
Electrochemical stability

At 700°C

R_p (as prepared) = $0.08 \Omega \cdot \text{cm}^2$

R_p (1 month / 700°C) = $0.08 \Omega \cdot \text{cm}^2$

OK R_p , stable



Very good performances

Promising stability

Bassat, J.-M.; Vibhu, V.; Nicollet, C.; Flura, A.; Fourcade, S.; Grenier, J.-C.; Rougier, A. *ECS Trans.* **2017**, *78* (1), 655–665.

<https://doi.org/10.1149/07801.0655ecst>

N. N. Greenwood, A. Earnshaw, "Chemistry of the elements", School of chemistry, University of Leeds, U.K., 1984

[Anderson, Don L.; 'Chemical Composition of the Mantle' in *Theory of the Earth*, pp. 147–175 ISBN 0865421234]

"Lanthanum". *price.metal.com*. *Shanghai Metals Market*. 3 February 2020. Archived from [the original](#) on 2020-02-03.

Federal Institute for Geosciences and Natural Resources. 22 January 2020. Archived (PDF) from the original on 2020-01-25.

Summary | $\text{La}_4\text{Ni}_3\text{O}_{10+\delta}$, $\text{Pr}_4\text{Ni}_3\text{O}_{10+\delta}$ & $\text{La}_3\text{PrNi}_3\text{O}_{10+\delta}$ investigation

Powder synthesis



- Le Bail & Rietveld refinement
- Crystalline structure with temperature

Oxygen content



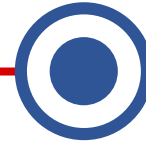
- δ -value
- Oxygen content with temperature

Conductivities



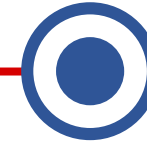
- Pellet densification
- Pulse isotopic exchange

Grain size tuning



- Attrition optimization

Cell preparation



- Tape test
- Preparation of a 3-electrodes setup

Powder synthesis

Tune grain size:
Attrition

Prepare ink

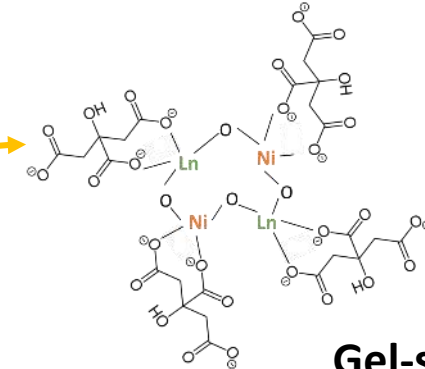
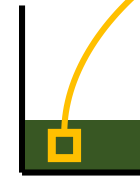
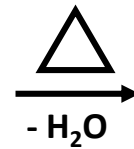
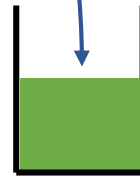
Screen-printing
deposition

« Sintering » to
The GDC substrate

Adhesion check
Tape test

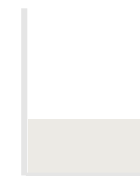
Modified Pechini method: Citrate route

Citric acid
 $\text{Ln}(\text{NO}_3)_3 \cdot x\text{H}_2\text{O}$
 $\text{Ni}(\text{NO}_3)_2 \cdot x\text{H}_2\text{O}$

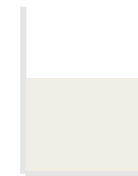


Gel-state

$\text{Ln}_4\text{Ni}_3\text{O}_{10+\delta}$



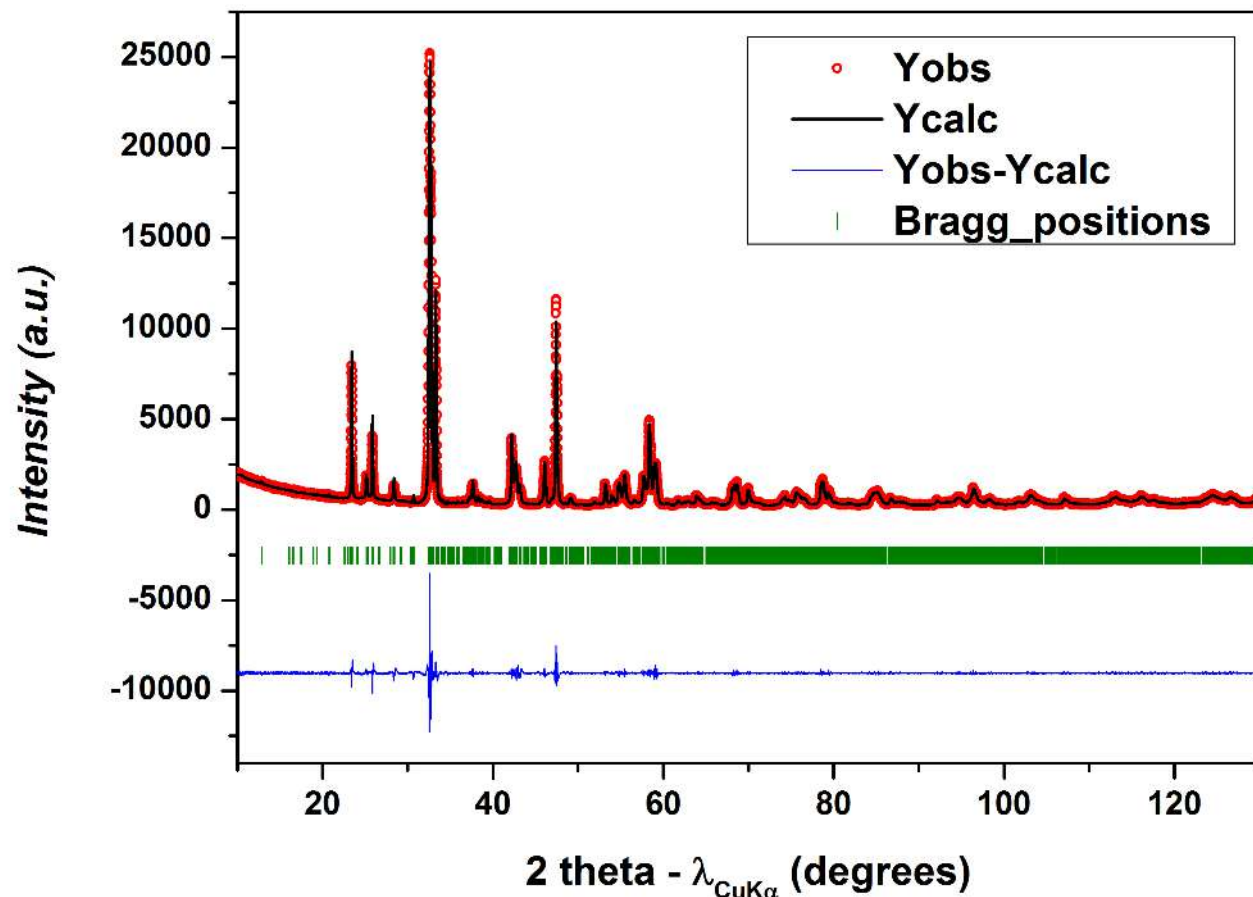
$\xleftarrow{1050^\circ\text{C}}$
 $\text{La}_4\text{Ni}_3\text{O}_{10+\delta}$: air
 $\text{Pr}_4\text{Ni}_3\text{O}_{10+\delta}$: O_2



$\xleftarrow{\text{Short Pyrolysis}}$

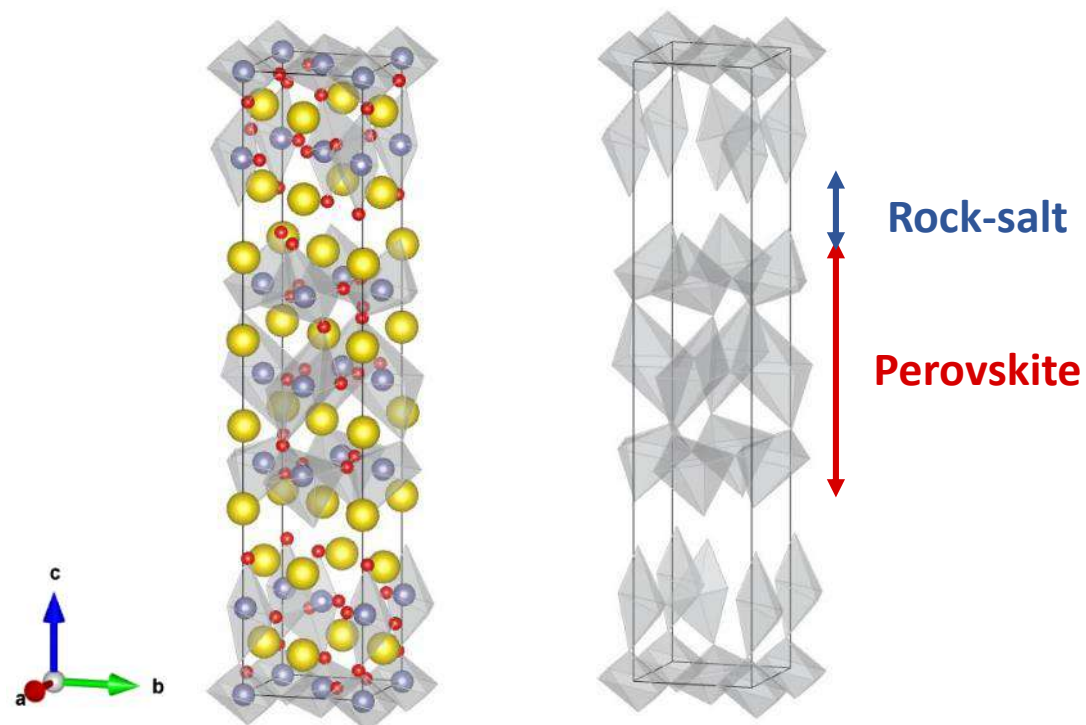


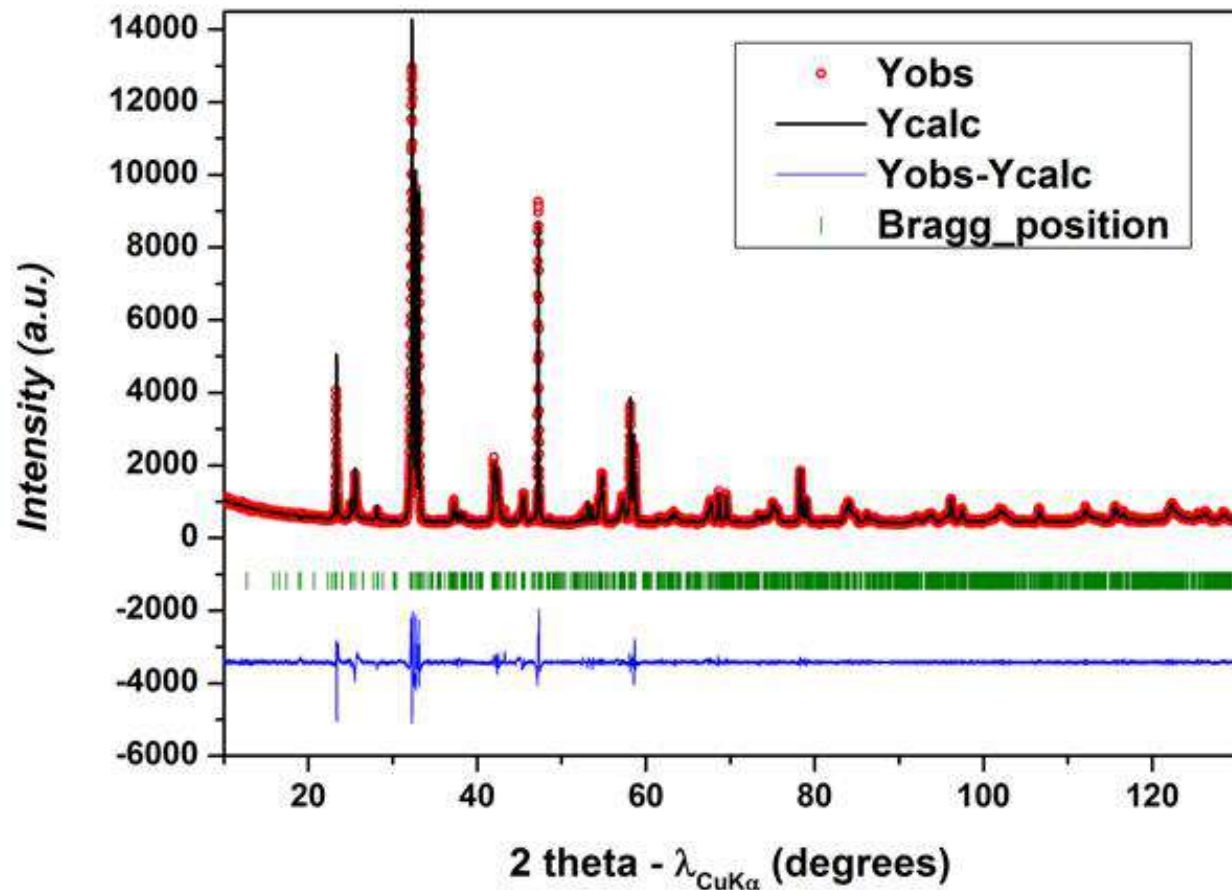
$\xleftarrow{\text{Heating On flame}}$



Cell parameter	Fitted	Literature
a (Å)	5.376469 (7)	5.37556 (5)
b (Å)	5.462346 (7)	5.46462 (6)
c (Å)	27.54941 (4)	27.4100 (3)
β (°)	90.29 (16)	90.283 (1)

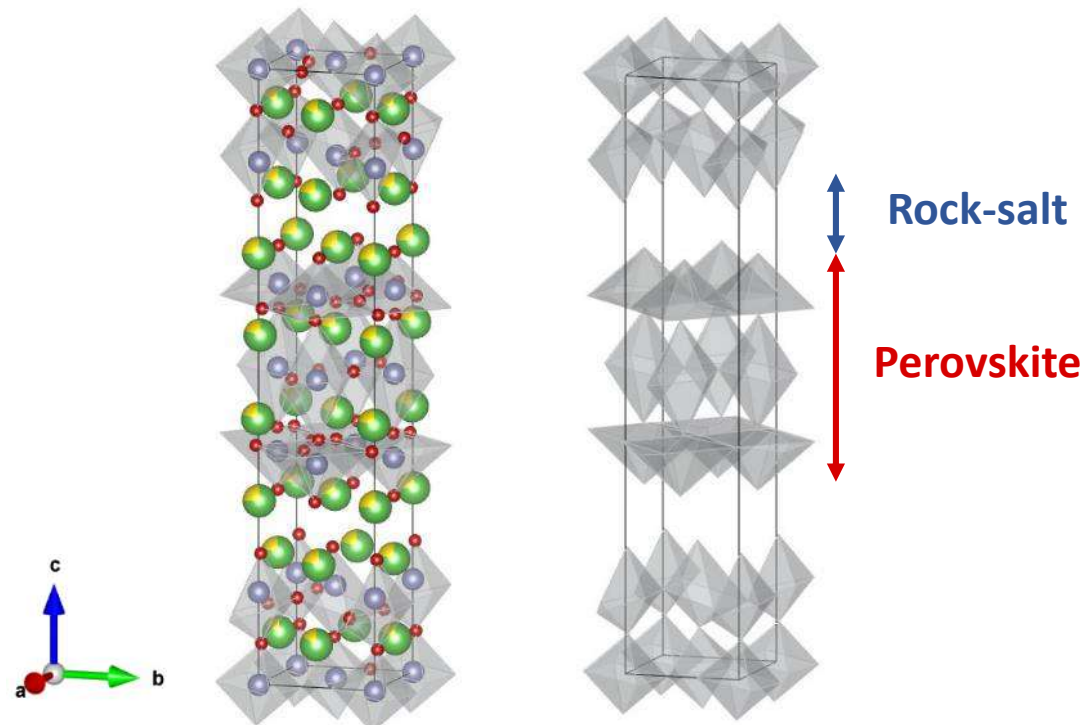
Monoclinic – $P2_1/a$

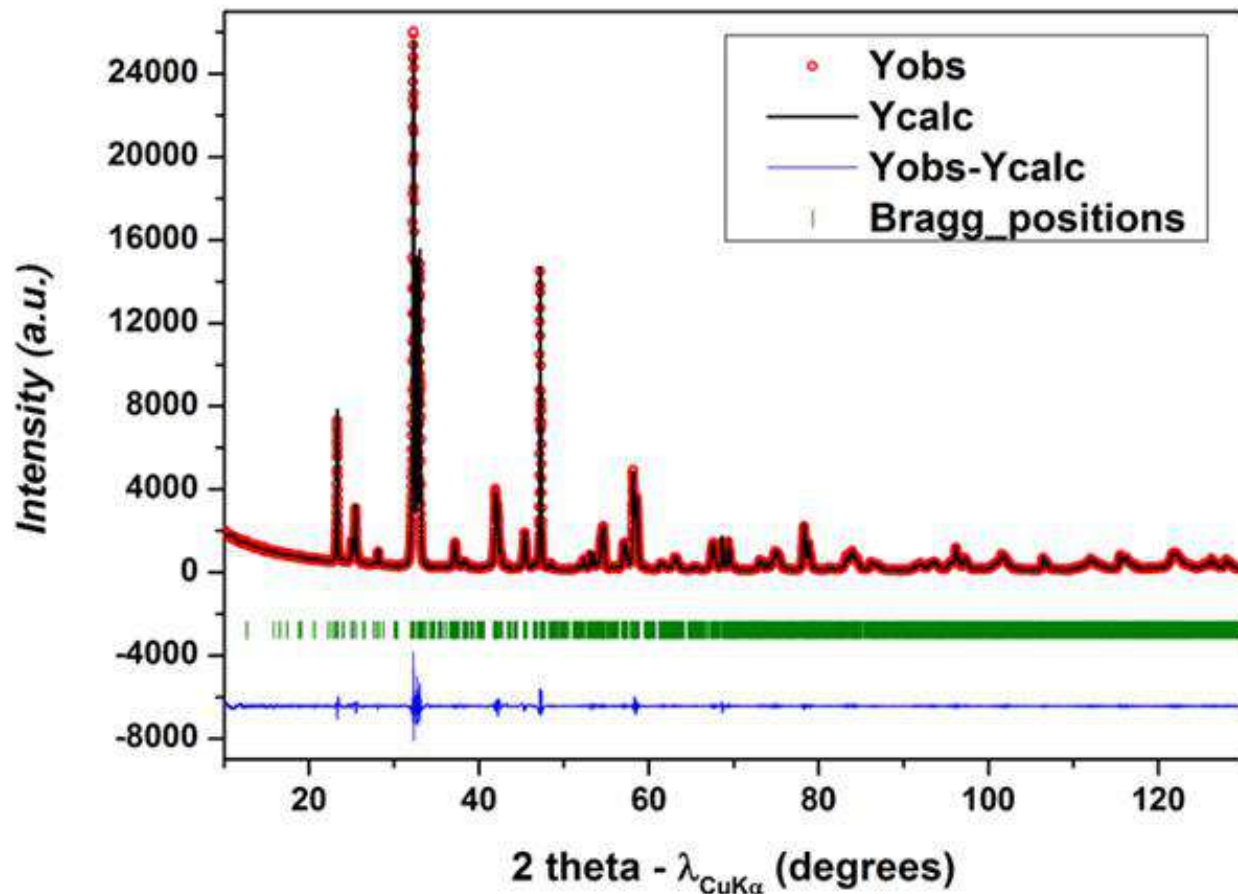




Cell parameter	Fitted	Literature (Fmmm)
a (Å)	5.40531 (8)	5.406 (1)
b (Å)	5.46614 (8)	5.466 (1)
c (Å)	27.889 (10)	27.873 (9)
β (°)	90.2562 (28)	90

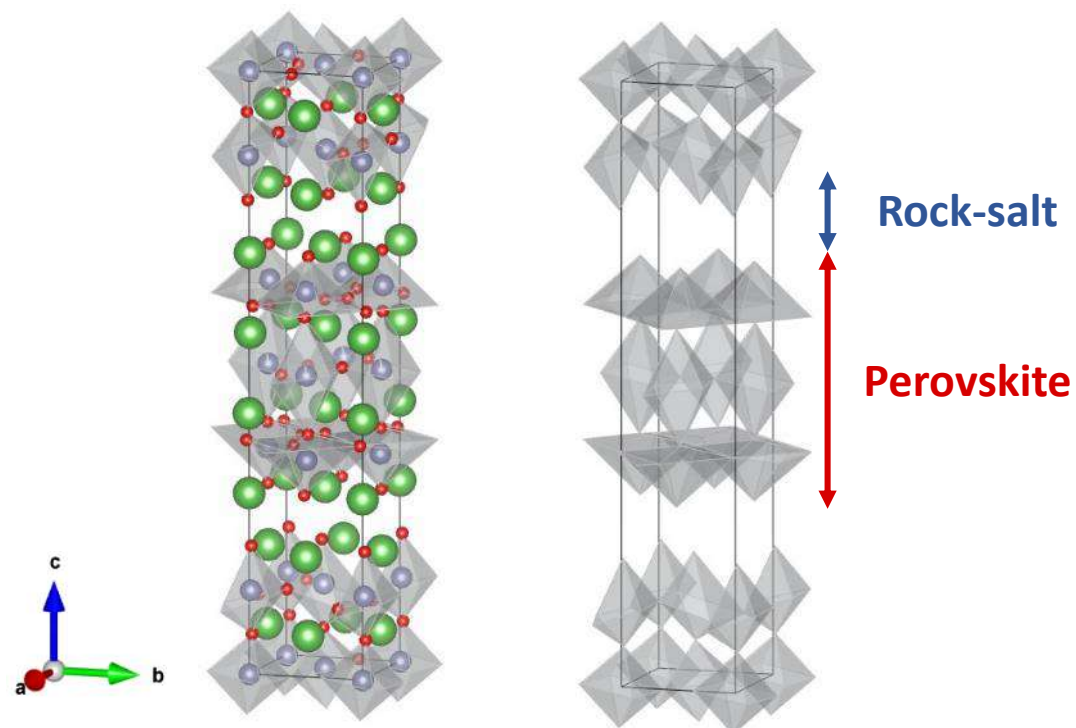
Monoclinic – $P2_1/a$

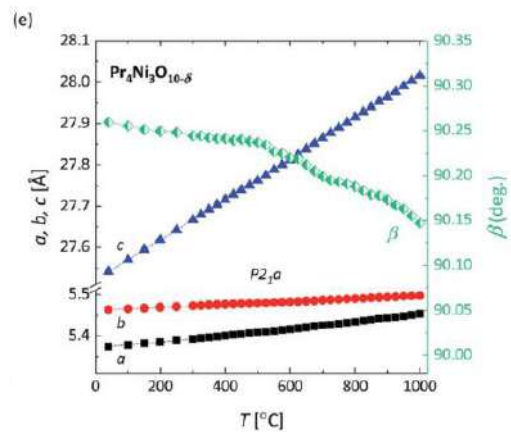
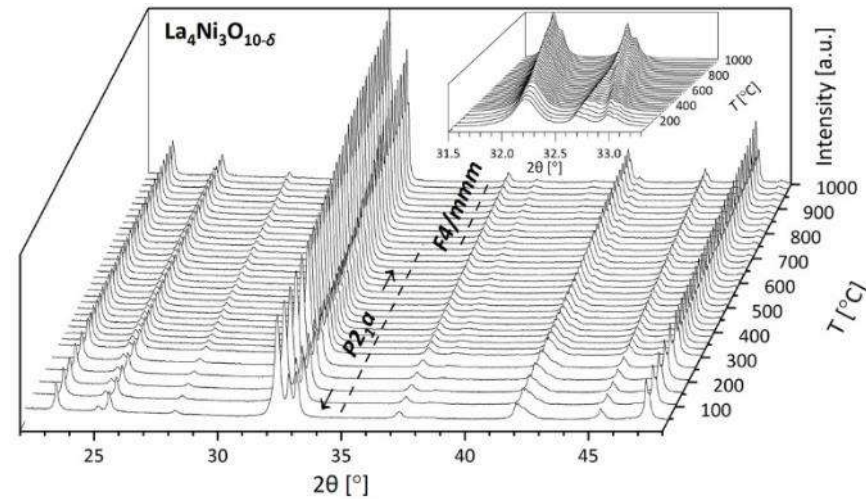
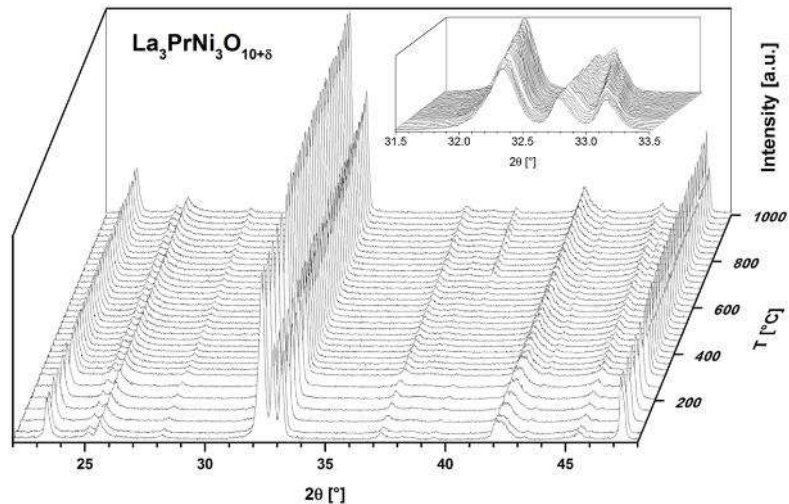
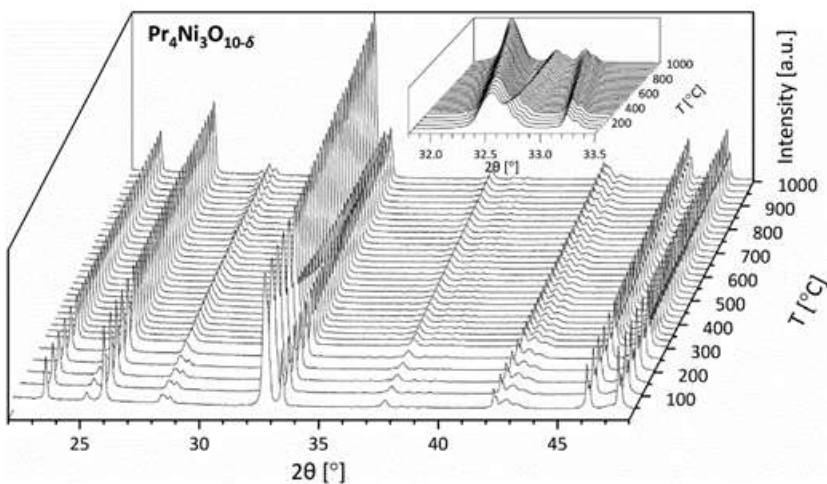




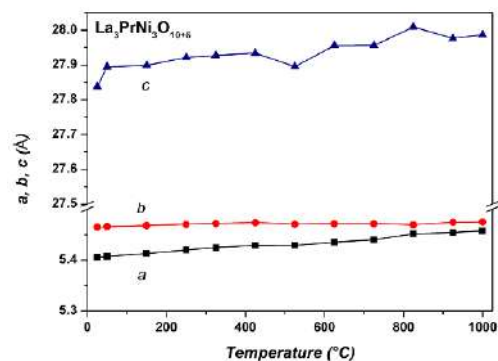
Cell parameter	Fitted	Literature
a (Å)	5.4166 (3)	5.37556Å
b (Å)	5.46544 (3)	5.46462Å
c (Å)	27.96155 (32)	27.4100Å
β (°)	90.198 (1)	90.283°

Monoclinic – $P2_1/a$

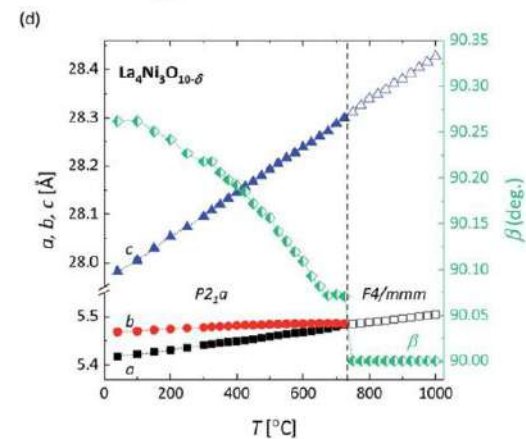




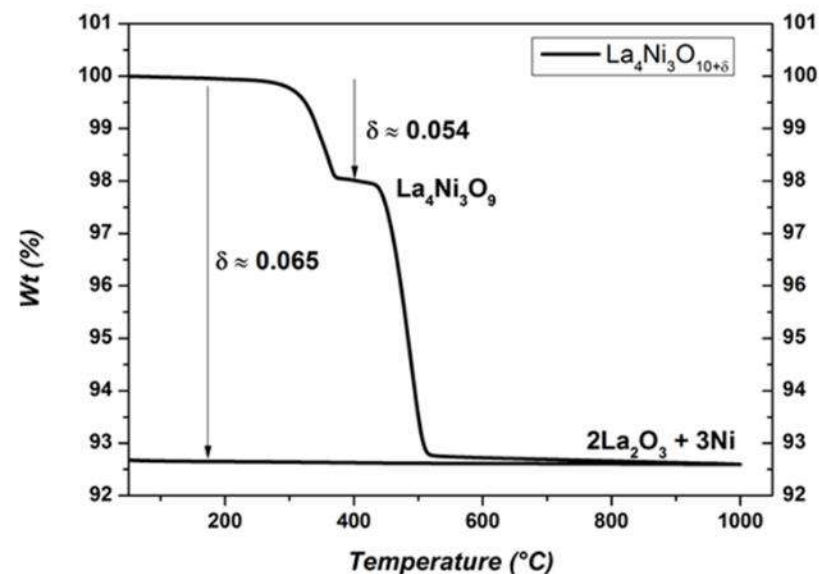
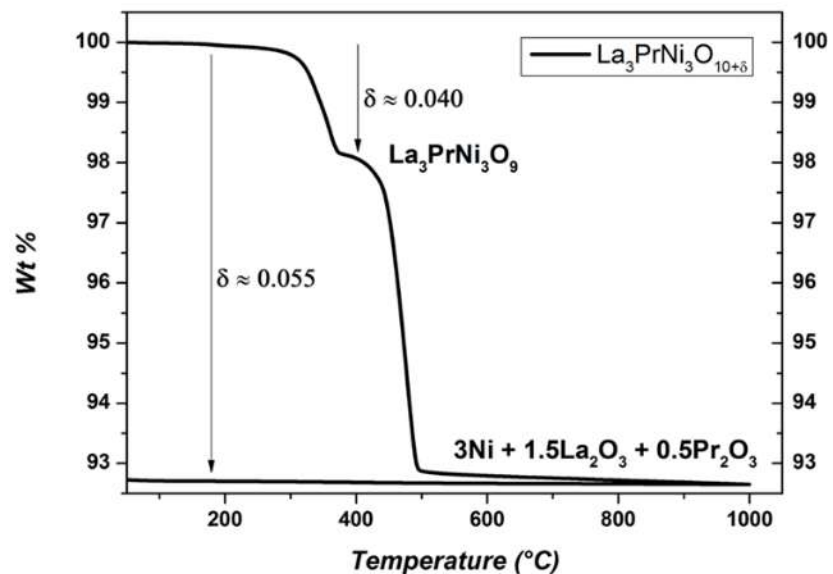
[Song *et al.*]



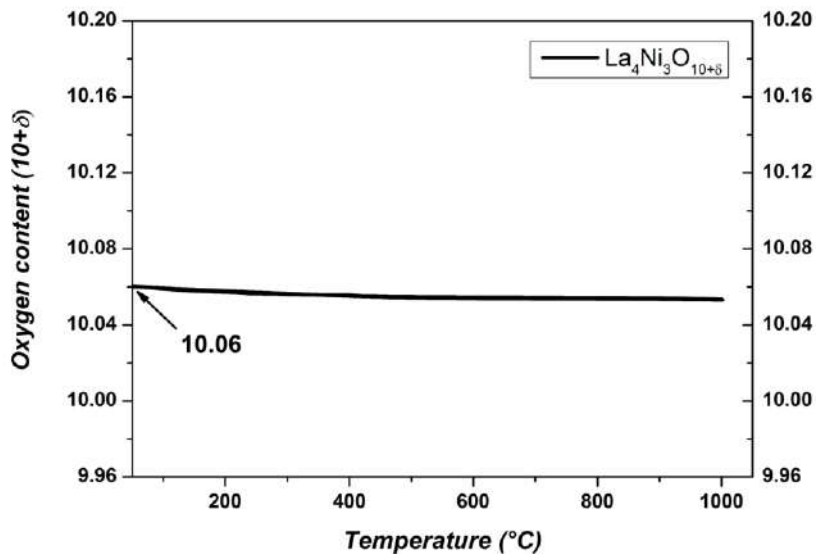
[This work]



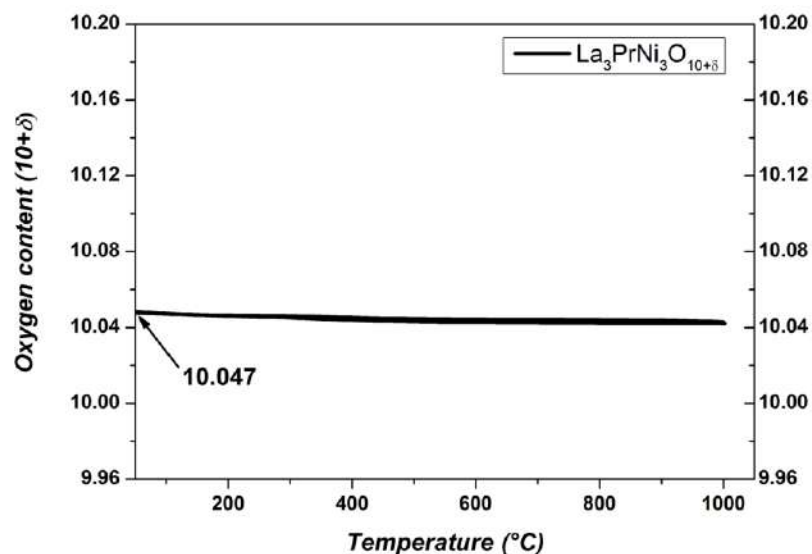
[Song *et al.*]



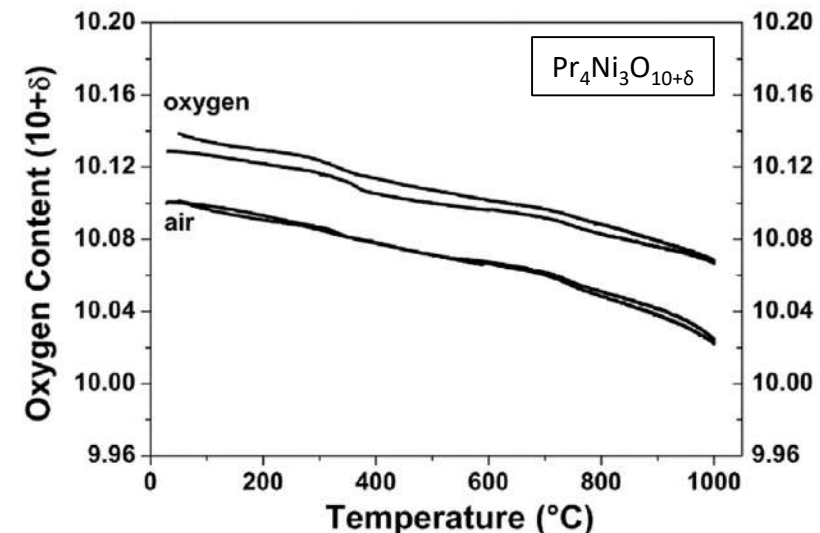
Material	δ -value	Source	Synthesis	Gas
$\text{Pr}_4\text{Ni}_3\text{O}_{10+\delta}$	-0.21 ± 0.03	Song. J <i>et al.</i>	1000°C/ 125h	O_2
	0.1 ± 0.01	Vibhu <i>et al.</i>	1000°C/ 48h	
$\text{La}_4\text{Ni}_3\text{O}_{10+\delta}$	0.06 ± 0.005	This work	1050°C/ 60h	Air
	-0.15 ± 0.04	Song. J <i>et al.</i>	1050°C/ 132h	
	0.06	Sharma <i>et al.</i>	1050°C/ 4h	
$\text{La}_3\text{PrNi}_3\text{O}_{10+\delta}$	0.047 ± 0.008	This work	1050°C/ 60h	Air



[This work]



[This work]



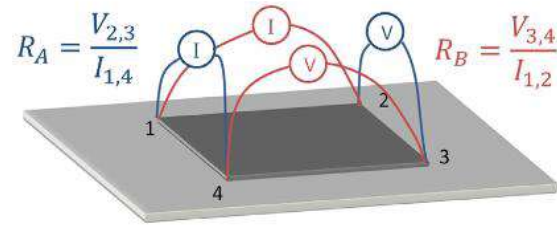
[Vibhu et al.]

Stable in air till 1000 $^{\circ}\text{C}$
Material « breathing » is observed in air

Electronic conductivity

4-probes characterization

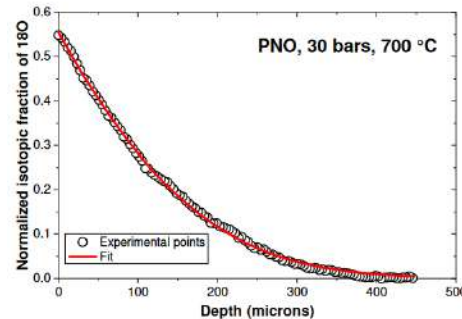
→ Need >96% dense pellet



Ionic conductivity

IEDP & ECR

→ Need >96% dense pellet



Anions diffusion pathways

Neutron diffraction

→ Possible on powder

→ Neutron diffraction beam time

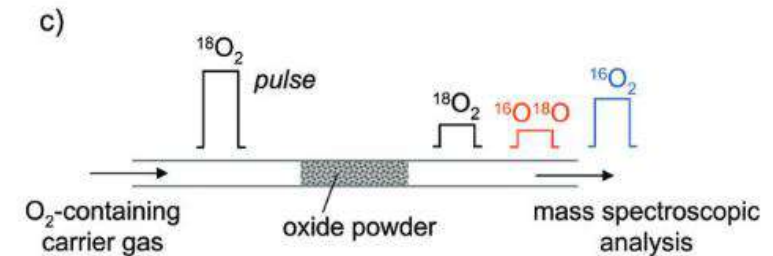
Granted june 2023



Oxygen exchange rates



Pulse isotopic exchange

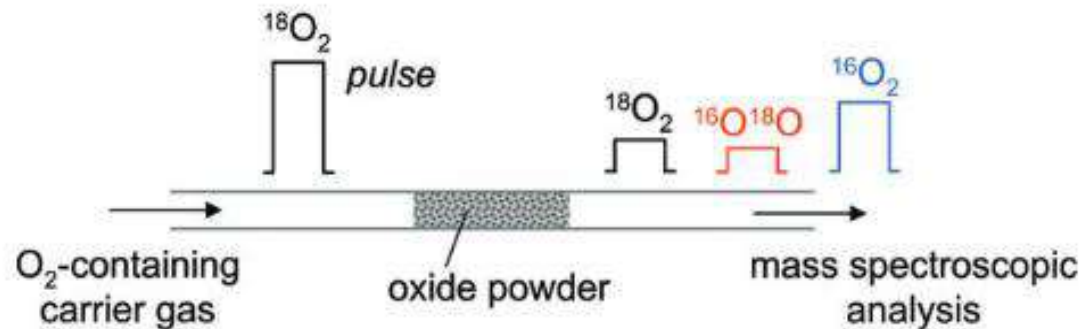
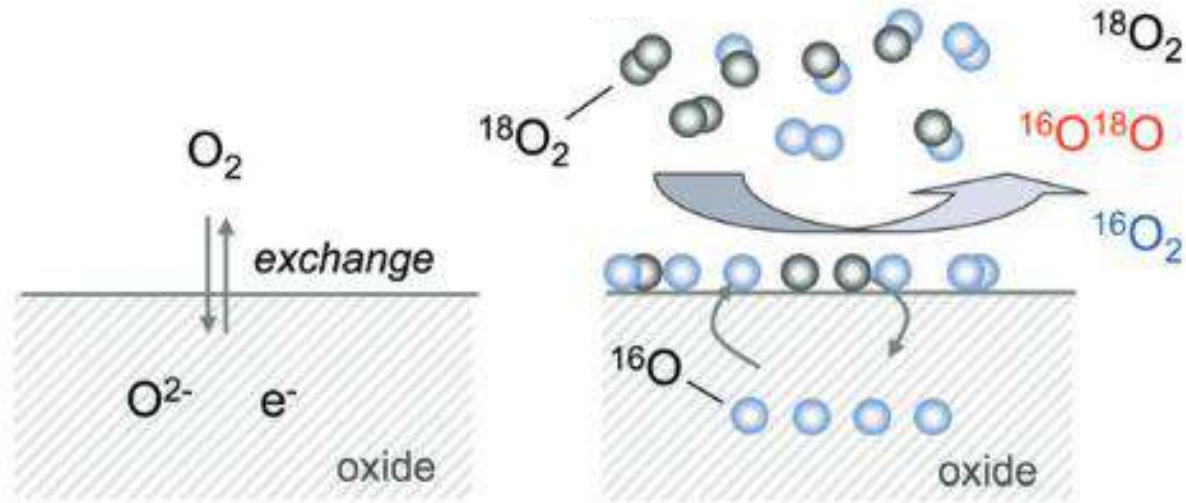
→ Possible on powder



Densification | Different strategies

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Material	Uniaxial press	Isostatic press	Annealing conditions	Relative density	Pellet
$\text{La}_4\text{Ni}_3\text{O}_{10+\delta}$	4MPa/ 5min	\emptyset	1050°C/ 350h/ air	72%	
$\text{La}_3\text{PrNi}_3\text{O}_{10+\delta}$	4MPa/ 5min	\emptyset	1050°C/ 350h/ air	82%	
$\text{La}_4\text{Ni}_3\text{O}_{10+\delta}$	4MPa/ 5min	300MPa/ 2min	1050°C/ 686h/ air - annealing in progress		
$\text{La}_3\text{PrNi}_3\text{O}_{10+\delta}$	4MPa/ 5min	300MPa/ 2min	1050°C/ 686h/ air - annealing in progress		
$\text{Pr}_4\text{Ni}_3\text{O}_{10+\delta}$	4MPa/ 5min	300MPa/ 2min	1050°C/ 512h/ O ₂ - annealing in progress		

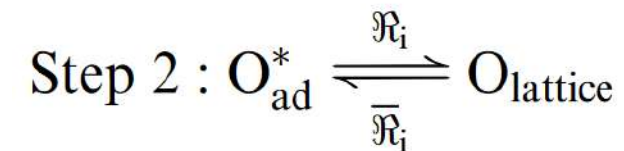
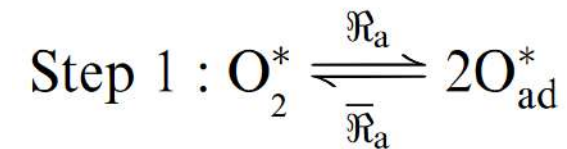


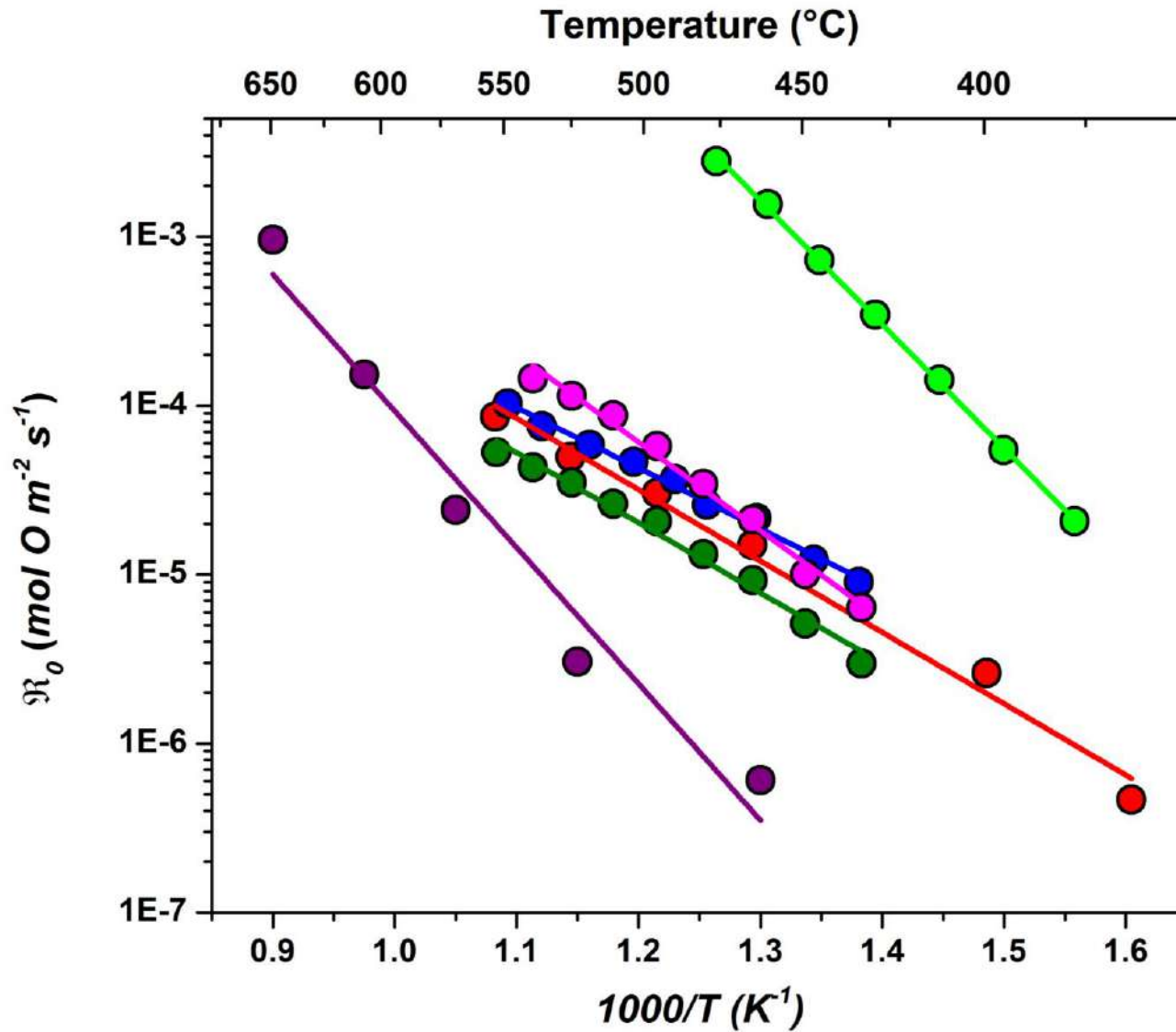
Allows reaching some parameters

R_0 : Balanced exchange rate under equilibrium conditions

R_a : Rate of dissociative adsorption

R_i : Rate of oxygen incorporation in the lattice





- BSCF Bouwmeester *et al.*
- $\text{La}_4\text{Ni}_3\text{O}_{10+\delta}$ Steinberger Wilckens *et al.*
- $\text{La}_4\text{Ni}_3\text{O}_{10+\delta}$ this work
- $\text{Pr}_4\text{Ni}_3\text{O}_{10+\delta}$ this work
- LSCF Gamon *et al.*
- LSCF Benson *et al.*

LSCF: reference SOCs air electrode material

BSCF: reference oxygen exchange material

[Majewski, A. J.; Khodimchuk, A.; Zakharov, D.; Porotnikova, N.; Ananyev, M.; Johnson, I. D.; Darr, J. A.; Slater, P. R.; Steinberger-Wilckens, R. Oxygen Surface Exchange Properties and Electrochemical Activity of Lanthanum Nickelates. *Journal of Solid State Chemistry* **2022**, *312*, 123228. <https://doi.org/10.1016/j.jssc.2022.123228>]

[Bouwmeester, H. J. M.; Song, C.; Zhu, J.; Yi, J.; Van Sint Annaland, M.; Boukamp, B. A. *Phys. Chem. Chem. Phys.* **2009**, *11* (42), 9640. <https://doi.org/10.1039/b912712g>]

Powder synthesis

**Tune grain size:
Attrition**

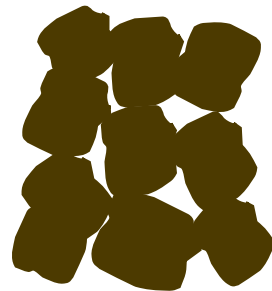
Prepare ink

Screen-printing
deposition

« Sintering » to
The GDC substrate

Adhesion check
Tape test

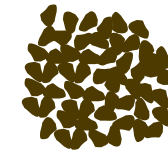
Fine powder = High surface area = Higher catalytic activity



As prepared powder

Grain size= 3.5 μ m

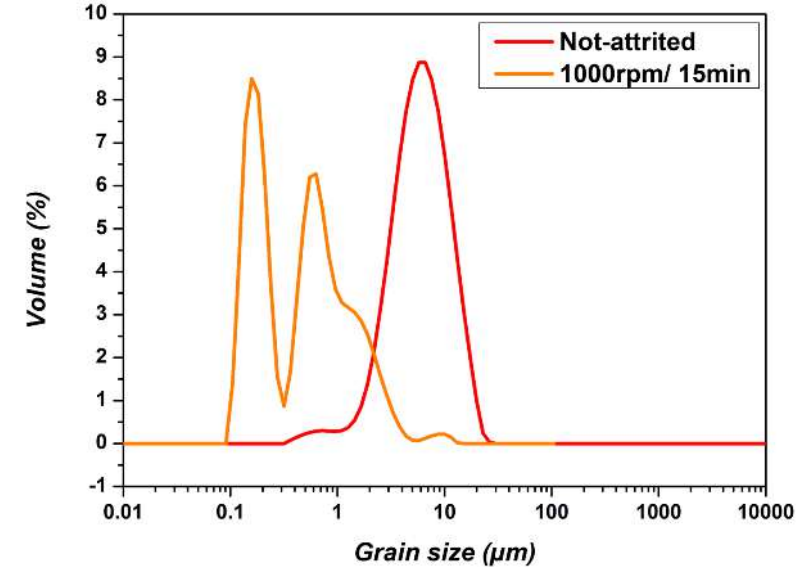
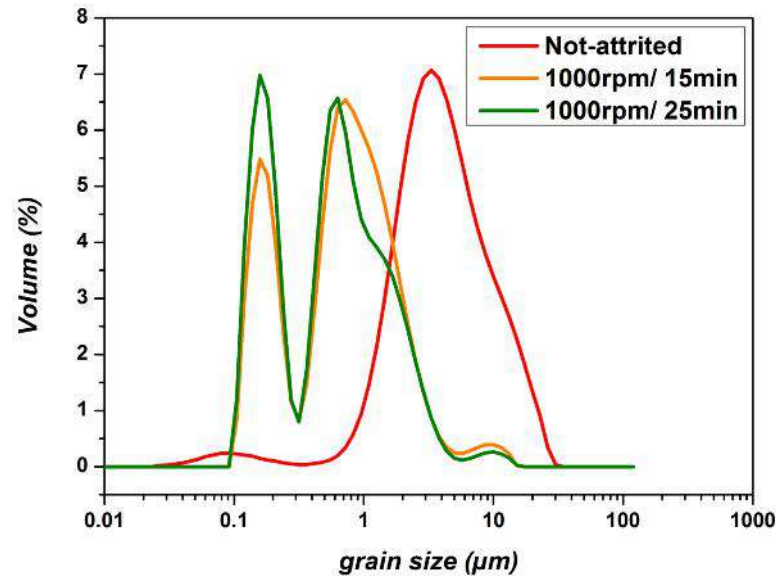
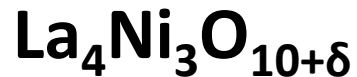
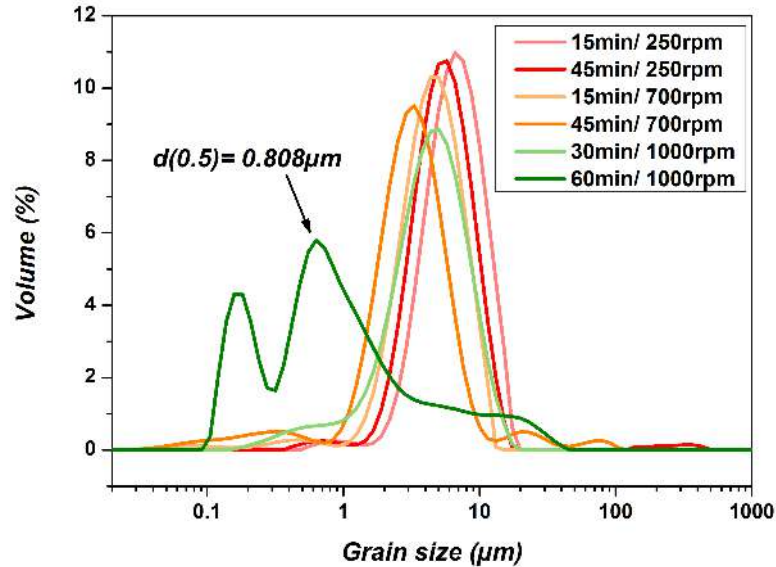
Attrition



Attrited powder

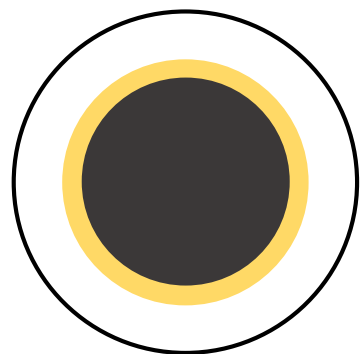
Grain size= 0.6-0.8 μ m

Aimed grain size: 0.6-0.8 μm according to Vibhu *et al.*



Post-attrition XRD: slight amorphization, no crystallites breaking

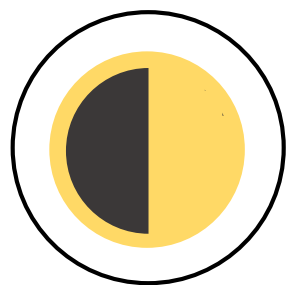
Aimed adhesion: The porous layer mustn't go off when pulling out a tape from it



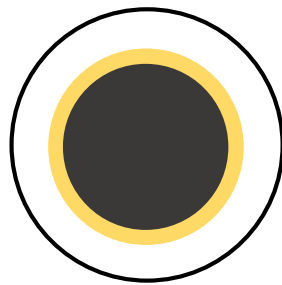
Electrolyte: 3YSZ

Barrier layer: CGO

Electrode: Nickelate



Poor adhesion



Good adhesion



950°C/ 2h



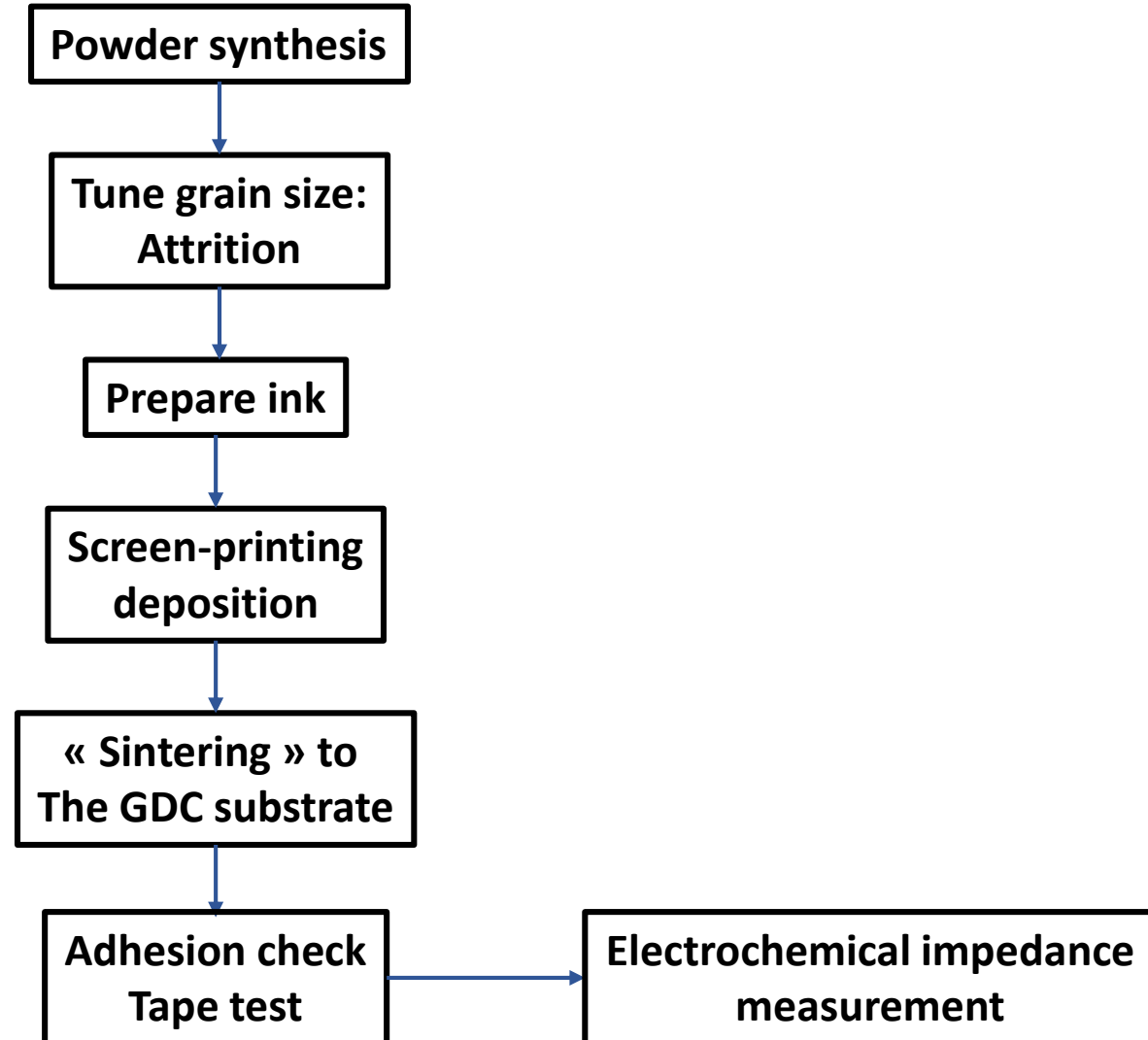
950°C/ 6h

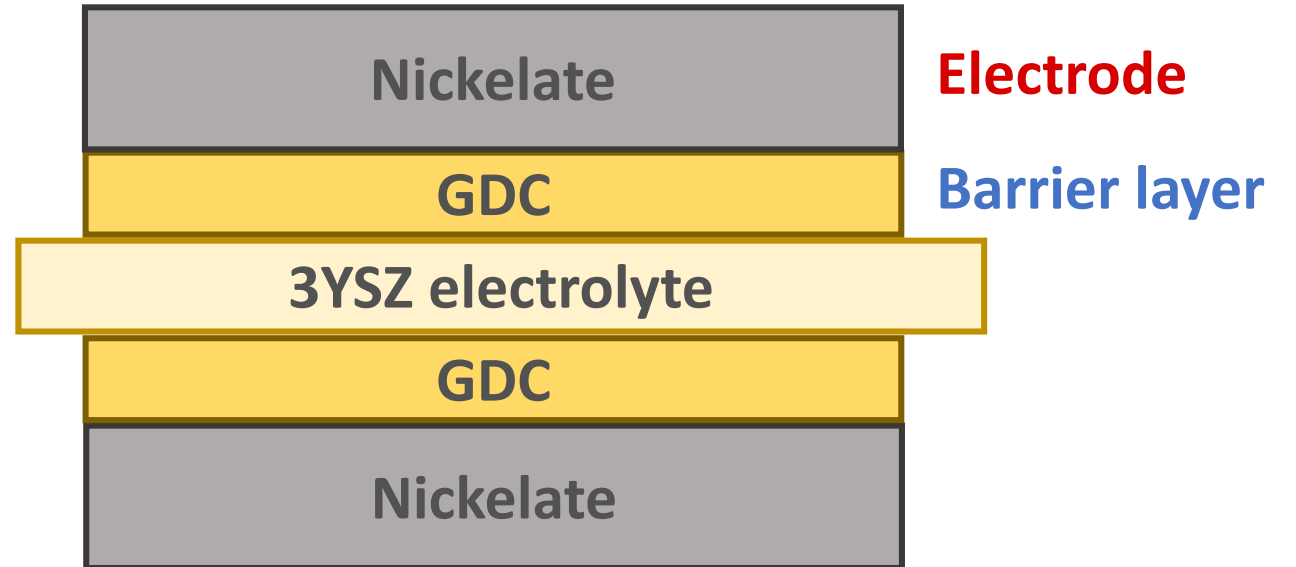
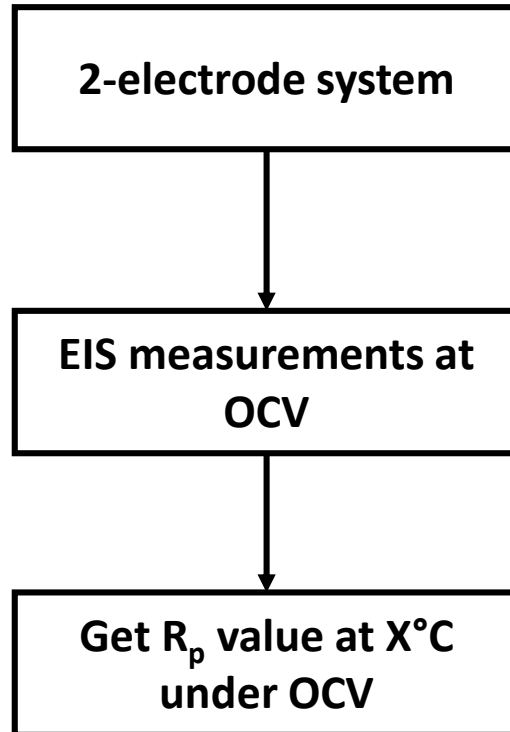


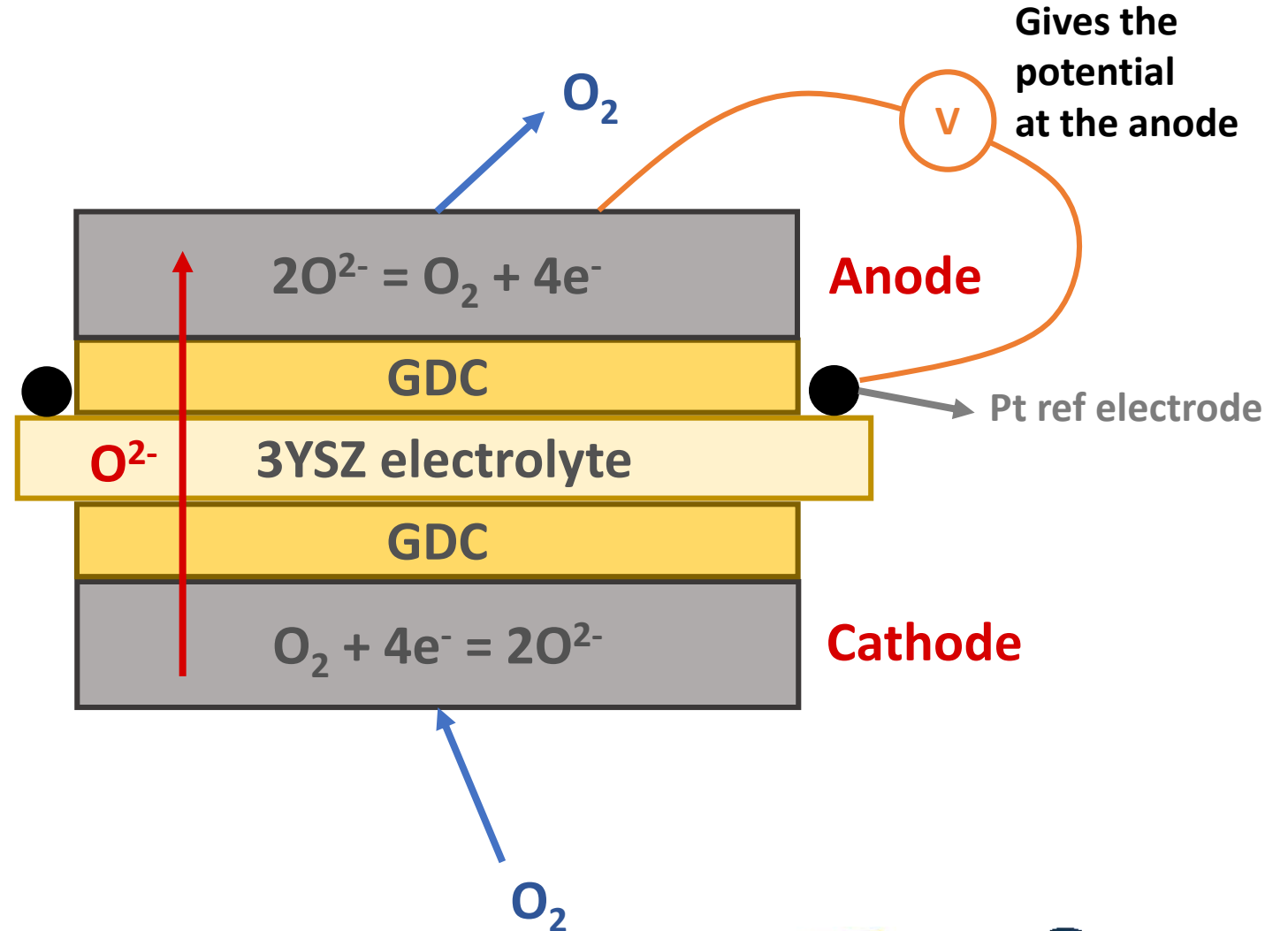
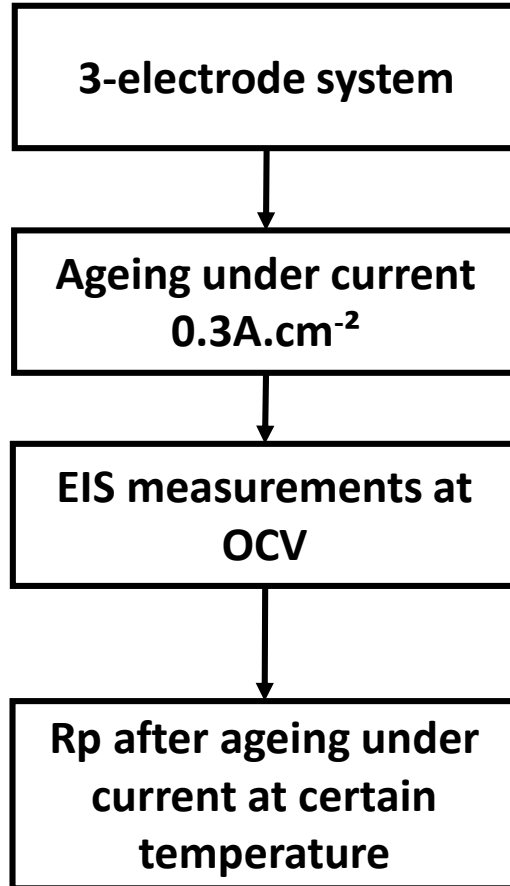
1050°C/ 2h



1050°C/ 6h

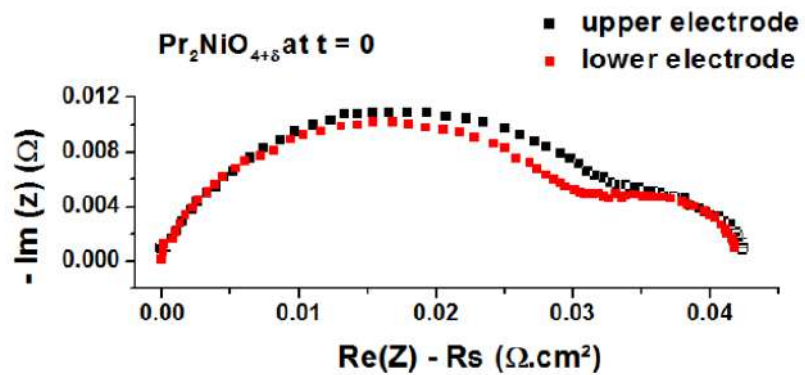






2-electrodes system

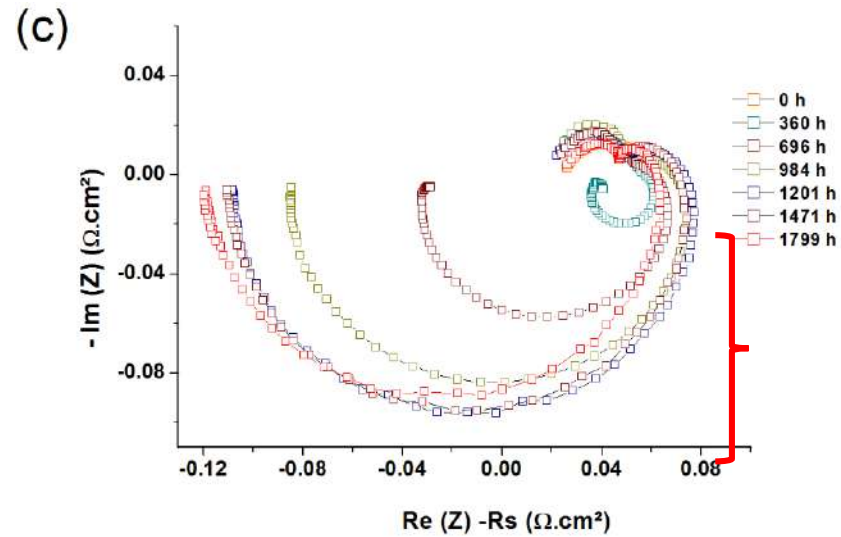
$i_{dc}=0$



Work on PNO

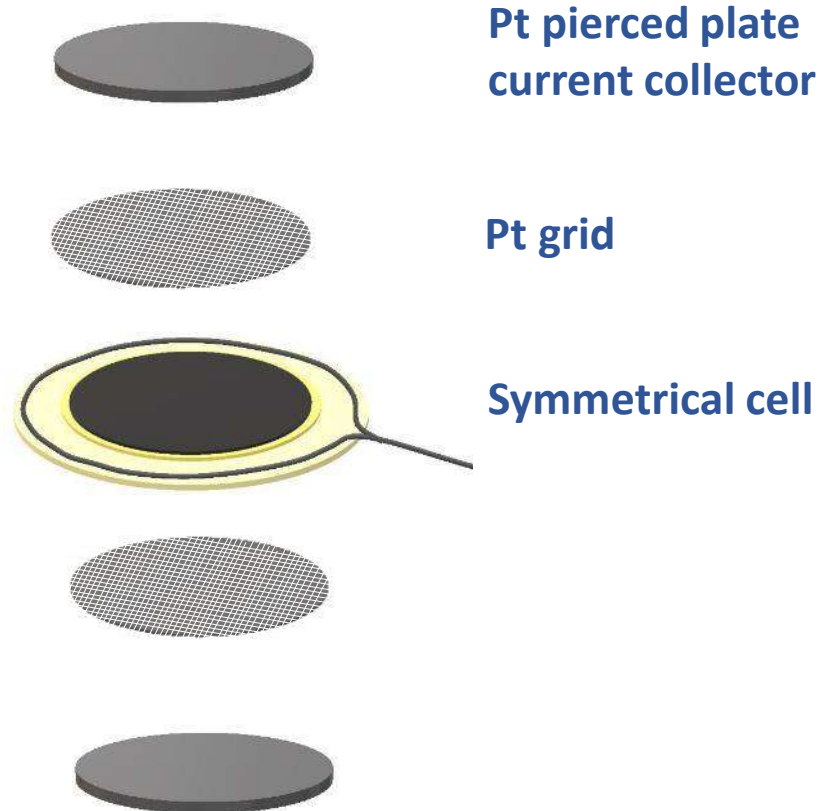
3-electrodes system

SOEC mode – $i_{dc} = -0.3 \text{ A.cm}^{-2}$



Large inductive loops
impeding the analysis

Set-up sketch

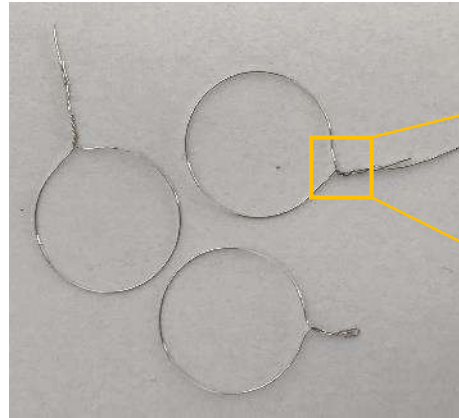


Setup under development

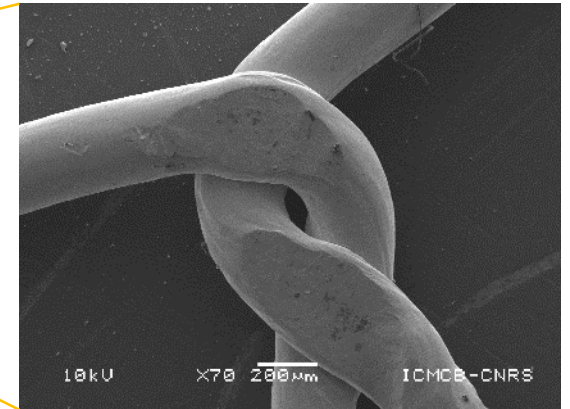
+ Screen-printer upgrades



Shaping really tricky!



Hard to get a perfect circle!



Welding by one spot welder



After several careful steps the 3 electrodes cell is eventually ready for EIS measurements

Structural & conductivities characterization

Neutrons diffraction

4-probes measurements

IEDP

ECR

Electrochemical measurements

EIS at OCV

Ageing under current

Complete cell characterization

Promising materials

Electrolysis assessment of $\text{La}_{1.5}\text{Pr}_{0.5}\text{NiO}_{4+\delta}$

Synthesis, characterization and, electrochemistry of $\text{La}_{2.25}\text{Pr}_{0.75}\text{Ni}_2\text{O}_{7+\delta}$

Acknowledgements

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PEPR CELCER-EHT program:



ICMCB & CNRS:



Supervisors:

Jacinthe Gamon & Jean-Marc Bassat



Co-supervisor:

Sebastien Fourcade



Workshop: 3-electrode system

Bertrand Guillaume & Olivier NGuyen

Co-author:

Vaibhav Vibhu



Collaboration for the PIE setup:

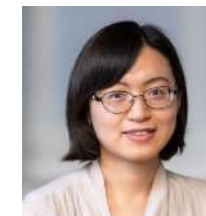
H.J.M. Bouwmeester

UNIVERSITY OF TWENTE.



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Thanks for your attention

Do not hesitate if you have any question!

